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Mudstones and embedded concretions show differences in lithology-related, but not source-related biomarker distributions

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Highlights

- Biomarker distributions of concretions and embedding mudstones were similar.
- Only lithology related indices differed between concretions and mudstones.
- Biomarker composition of mud is preserved during concretion formation.
- Biomarker composition of Gogo nodules is related to the palaeoenvironment at time of deposition.

- 1 Mudstones and embedded concretions show differences in lithology-related, but not source-
- 2 related biomarker distributions
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- 11

12 Abstract

13 The mudstones of the Western Australian Gogo Formation harbour numerous carbonate concretions which often contain preserved fossils of Lagerstätte-like quality. These are especially notable in 14 15 places where the mudstone has eroded, giving way to nodule fields, where they are collected for the 16 paleobiological records of great interest they afford. It is, however, a challenge to determine their paleoenvironmental context. Here, we analysed concretions from a core drilled in the Canning Basin 17 Gogo Formation. At two different depths, concretions were compared to the surrounding mudstone 18 19 found at the same depth. Electron microscopy and X-ray spectroscopy showed that the concretions 20 were carbonate-rich and contained detrital fragments. Biomarker data showed that mudstone and 21 concretions were very similar and presented marine biosignatures including indicators of anoxic depositional conditions, a stratified water column, and photic zone euxinia. The concretions contained 22 higher amounts of C₂₇ steranes, indicating that more labile organic matter such as animal remains 23 24 could have triggered concretion formation. Statistical analyses of the results showed that concretions 25 and shales differed (p < 0.1) in indices diagnostic for lithologies (and often related to maturity),

26 particularly in the sample recovered from the younger section of the core: the diasterane/sterane 27 indices at each of the depths were 0.34 in the shale vs 0.12 in the concretion in the younger, and 0.2128 vs 0.19 in the older set of samples. The homohopane isomerisation indices of mudstone and 29 concretion were 0.66 vs 0.58 and 0.62 vs 0.60 (47 and 54 m depth, respectively). Further, shales 30 contained higher relative amounts of moretanes. Concretions also differed from the mudstone in their 31 methylhopane distributions, with shales showing higher amounts of 3-methylhopanes. Conversely, 32 when biomarker composition was compared at the two depths, they only differed significantly (p < p33 0.1) in 2-Methylhopane index, % C_{31} Homohopanes, and hopane/sterane ratios (p < 0.1). Our results 34 show that the biomarker composition of the mudstone at the time of deposition is largely preserved in the concretions with only a minor overprint from diagenesis, bacterial communities involved in 35 36 concretion formation, or the nucleus. It is therefore possible to use biomarker analysis on carbonate 37 concretions to determine their provenance if found outside their immediate geological context.

38

39 1. Introduction

The vast stromatoporoid reefs of the Devonian Period (419.2-358.9 Ma) harboured complex and 40 41 diverse ecosystems, to which formations such as those found in the Canning Basin in the North of 42 Western Australia bear witness (Playford, 1980; Playford et al., 2009). The Gogo Formation is part of 43 this fossil reef complex, and composed of basin facies stemming from the Frasnian to the Givetian. In 44 the Gogo mudstone, the presence of carbonate concretions is notable, and preserved fossils of high quality, close to that of a Lagerstätte, can be found (Long and Trinajstic, 2010). Carbonate concretions 45 46 have formed diagenetically since the Proterozoic, and are often associated with organic matter-rich 47 sediments (shales, mudstones) and sandstones. It is generally accepted that deposition of organic 48 matter in low oxygen conditions supports their growth and preservation (Berner, 1968; Allison, 1988; Briggs and Kear, 1993; Briggs, 2003; Long and Trinajstic, 2010). They often form as a result of 49 50 bacterial activity in the anoxic muds, and grow concentrically or pervasively around a centre of 51 decaying organic matter, where microbial sulfate reduction and methanogenesis cause changes in 52 alkalinity (Raiswell and Fisher, 2000). These changes result in fast precipitation of

53 carbonatesincluding calcite and dolomite, as cements in the pore spaces of unconsolidated sediment, 54 and preservation of the original characteristics of the organic matter nucleus (Wolff et al., 1992). A 55 large number of studies have shown excellent morphological fossil preservation in concretions 56 spanning the entire Phanerozoic (e.g. Weeks, 1953; Marshall and Pirrie, 2013; Wilson and Brett, 57 2013). Many of these concretions have been eroded out of their embedding rock, are found as large 58 nodules fields in the Canning Basin, and their enclosed fossils, such as the Gogo fish, have afforded 59 major paleobiological insights (Long and Trinajstic, 2010). Biomarker analyses of concretions has 60 shown the preservation of biolipids in very immature stages (Kiriakoulakis et al., 2000; Marynowski 61 et al., 2008; Melendez et al., 2013a, 2013b). Melendez et al. (2013b, 2013a) described the biomarker 62 composition of a soft-tissue concretion from the Gogo Fm. and found exceptionally preserved 63 biomarkers and biomolecules, including a diagenetic continuum of steroids. However, it is poorly 64 understood how the biomarker composition of the Gogo nodules relate to that of the mudstone they 65 are immediately embedded in. A recent comparison of concretions from the Toarcian with their 66 surrounding shale (Posidonia shale, SW German Basin) showed that the biomarker composition of 67 concretions was similar to those of the surrounding shales and thus to environmental conditions at 68 time of formation in those pyritic, organic matter (OM) containing carbonate nodules from this 69 location (Plet et al., 2016). However, this hypothesis requires testing in more settings. Here, we investigated the biomarker compositions of concretions from a core from the Gogo Formation and 70 71 compared it to the embedding mudstone. The results could allow the use of the biomarker 72 composition to place eroded nodules, such as those found in the North Western Australian nodule fields, in a confirmed paleoenvironmental context. 73

74

75 2. Methods

76 2.1. Geological setting and core

A core drilled at 18.4818°S 125.9768°E in 1983 by the East Pillara Joint Venture stored at the core
repository of the Western Australian Geological Survey (ED20) was examined (Benn and Styles,
1984). It stems from the Canning Basin and was in total 500 m long, of which the upper 110 m

spanned the Gogo Fm. (basin facies of the Givetian to Frasnian; Playford, 1980) consisting of
laminated, alternating light and dark grey mudstone with little fauna and of low thermal maturity
(Schwark, 2013). At two depths, 47.15–47.45 m, and at 54.56–54.86 m depth, carbonate concretions
were found embedded in the mudstones; both were sampled for geochemical analysis. The concretion
at 54 m depth was large and partially cored (Fig. S3), whereas the concretion at 47 m depth was a
small round nodule of < 10 cm diameter.

86 2.2 Scanning electron microscopy

87 The small concretion, taken from the core section at 47 m depth, was cut, polished and coated with Pt 88 using a Cressington 208HR. Scanning electron microscopy and energy-dispersive X-ray spectroscopy 89 imaging was carried out with a Tescan Mira3 FESEM and an Oxford Instruments X-Max 150mm² 90 and AZtec software at 15 mm working distance and 20 kV. X-ray diffraction (XRD) patterns of the powdered shales were collected on a D8 Advance diffractometer (Bruker AXS, Germany) using Cu 91 92 Ka radiation. The data were collected using a 20 step size of 0.015° , a count time of 0.5 s per step and 93 a 2 θ range of 5°–120°. Rietveld modelling with the data was performed using Bruker AXS TOPAS 94 version 5. A corundum (Al₂O₃) internal standard (10 wt.%) was used to facilitate quantitative 95 analysis.

96 2.3 Extraction, fractionation and derivatisation

97 The concretions and mudstones were cleaned by 5 x 15 min sonication in dichloromethane /methanol 98 1:1 (v:v), followed by grinding with a rock grinder using a ceramic bowl that had been baked at 450 99 °C. The resulting powder was extracted via Soxhlet in dichloromethane (DCM) / methanol (MeOH) 9:1 (v:v) for 72 hours. The extract was concentrated by rotary evaporation and weighed. Aliquots of 100 the total lipid extracts were fractionated by elution over activated silica (SiO₂) with 3 column volumes 101 102 of hexane (saturate fraction), 3 column volumes of hexane/DCM (7:3, v:v, aromatic fraction) and 3 103 column volumes of DCM/MeOH (1:1, v:v, polar fraction). For the concretions, the total lipid extract was used after derivatisation with N₀-Bis(trifluoro)acetamide (BSTFA) in pyridine for 1 h at 70 °C. 104 105 A procedural blank of annealed sand was also processed in the same way. Only a few contaminants,

predominantly plasticizers, were identified in the blank. As the focus of this study was to obtain and describe the preserved biomarker fraction rather than determining the fatty acid composition of the concretions, HCl dissolution prior to extraction (Wolff et al., 1992; Pearson et al., 2005) was in this case not employed.

110 2.4 Biomarker identification

111 The aromatic fraction of the mudstones, the total lipid extract (TLE) of the concretions and the procedural blank were analysed by gas-chromatography mass spectrometry (GC-MS) as described by 112 Tulipani et al. (2015a) in order to evaluate the carotenoid pigment contents. The saturate fraction of 113 114 the mudstones and the TLE of the concretions was analysed by gas-chromatography metastable 115 reaction monitoring (GC-MRM) (Tulipani et al., 2015b) in order to determine the sterane and hopane biomarkers and ratios and compared to a Blina oil standard. Some C33 3-methylhopanes were 116 tentatively identified by their retention behaviour according to Summons and Jahnke (1992). 117 118 Additionally, as the biomarker concentrations in the concretions were low, they were analysed by GC-119 MRM using a DB-5 MS column, monitoring the transitions m/z 554 \rightarrow 134 and m/z 546 \rightarrow 134 and compared to Blina oil and oil from the Perth Basin in order to identify carotenoid pigments (French et 120 al., 2015). Biomarker ratios were calculated from the areas of the MRM transition peaks as outlined in 121 Peters et al. (2004). Statistical analysis on biomarker ratios was conducted in R using the FactoMineR 122 123 package (Le et al., 2008).

124

125 3 Results and discussion

126 3.1. Inorganic Geochemistry

QXRD analyses revealed similar mineralogy of both mudstones, which largely consisted of quartz
and Muscovite (22 and 26%, Clay), and contained smaller amounts of brushite, pyrite and orthoclase.
Electron microscopy of the concretion found at 47 m showed the heterogeneous structure of the
concretion (Fig. 1). Elemental mapping confirmed that the main cement of the concretions was
CaCO₃ (Fig. 1B, Site 1), with fragments of SiO₂ (Fig. 1B, Site 2) and other silicates (Fig. 1B, Site 3)

132 such as K-Feldspar (Fig. 1B, Site 4) and pyrite (Fig. 1B, Site 5). Maps of individual elements of an 133 area containing a small vein (Fig. 1C) confirmed that it consisted of a silicate (Si, Al, O, C, Na, Fig. 1D) and also showed the presence of pyrite (Fe, S, possibly framboidal, Fig. 1D), fragments of apatite 134 135 (P, Fig. 1D) and albite (Na, Fig. 1D). No obvious preserved fossils or OM-rich areas were detected, 136 suggesting that the concretions were formed around a very small organic nucleus, or that the nucleus 137 had been completely replaced. The lithified carbonate in concretions stems from micro-138 environmental changes in pH and Ca or Mg ion concentrations by the microbial community involved 139 in organic matter degradation (Wright and Oren, 2005), while the presence of pyrite is indicative of 140 microbial sulphate reduction (MSR). The silicate fragments are probably detrital (Sugitani et al., 141 1995), although authigenic K-feldspar in carbonates has been reported previously and its formation 142 attributed to sulfate reduction coupled to anaerobic methane oxidation (Jørgensen, 1981). This could 143 provide an explanation for the appearance of silica-filled cracks (Fig. 1D), which could have formed 144 during compression of the carbonate concretions during burial and filled with authigenic silica by either inorganic or microbially-mediated reactions. However, if these cracks were septarian fractures, 145 146 these would normally be calcite-filled (Kiriakoulakis et al., 2000), and the features found here could rather present the remains of siliceous spicules of marine hexactinellids odemosponges (Uriz et al., 147 148 2003). These results confirm that the samples analysed were indeed post-depositional carbonate concretions. 149

150 *3.2 Biomarkers and derived indicators*

Pristane/phytane ratios were low for all samples, and amounted to 0.43 for the concretions and 0.47 151 and 0.57 for the mudstone, indicating exceptionally low thermal maturity for material of this age. The 152 153 aromatic fractions of the mudstones possessed small amounts of paleorenieratene and isorenieratene (Fig. 2AB), and GC-MRM analysis of the TLEs confirmed that they were also present in the 154 concretions (Fig. 2CD), providing evidence for photic zone euxinia during deposition of the sediment 155 156 (Grice et al., 1996; Sinninghe Damsté et al., 2001; Whiteside and Grice, 2016). All samples displayed a high proportion of C₂₉ steranes (Table 1, Fig. 3B, Supplementary. Fig. 1), which is common for 157 marine sediments older than 350yr (Moldowan et al., 1985), and would thus be expected from 158

159 Devonian reef basin facies such as the Gogo Fm. However, the concretions contained slightly more C_{27} steranes (Fig. 3B) and C_{27} diasteranes, while the mudstones contained more C_{29} steranes, 160 161 indicating that part of the sterane complement contained in the concretions could stem from decaying 162 animal remains providing labile organic matter acting as a nucleus for concretion formation. C_{27} 163 sterols, such as cholesterol, are typically present in higher abundances in animals. This observation 164 has recently also been made for Toarcian nodules in which the C_{27}/C_{29} sterane ratio was slightly 165 higher for the concretions than for the shales (Plet et al., 2016). Only small amounts of C₃₀ steranes 166 (*n*- and *iso*propylcholestane) were detected. C₃₀ steranes, though present in higher abundances in e.g. 167 the Cryogenian, have been reported to be scarce in samples from e.g. the Lower Paleozoic (Moldowan, 1984; Rohrssen et al., 2015; Gold et al., 2016). However, they are often present in 168 169 Devonian settings, albeit sometimes in equally low relative proportions (e.g. Tulipani et al., 2015a). 170 The fractional abundances of all $(17\alpha, 21\beta + 17\beta, 21\alpha + 17\beta, 21\beta)$ C₃₁₋₃₅ homohopanes, often used for fingerprinting, were very similar across all samples (Fig. 3A), with significant amounts of C₃₅ 171 homohopane, the presence of which is generally associated with sulfidic depositional conditions 172 173 (Sinninghe Damsté et al., 1989; Peters and Moldowan, 1991; Sinninghe Damsté et al., 1995). The 174 similarity of the homohopane fingerprints between muds and concretions indicates that a bacterial community involved in concretion formation did either not produce extended hopanes different from 175 176 the bacteria in the muds or overlying water column, or that the bacterial community was the same as 177 in the surrounding sediment, and only activity was enhanced due to a localised high concentration of labile organic matter. It is possible that preservation of the extended hopanes occurred by 178 incorporation into kerogen, during early diagenesis in the presence of reduced-sulfur compounds 179 (Kohnen et al., 1991; Sinninghe Damsté et al., 1995; Wakeham et al., 1995; Grice et al., 1998). 180 181 However, due to the small size of the concretions (< 10 cm) as sampled from the core, desulfurisation 182 and hydrogen pyrolysis to analyse the sulfur-bound and kerogen-bound fractions could not be applied. C_{29}/C_{30} hopane ratios were highest in the mudstone at 47 m depth, slightly lower in the corresponding 183 184 concretion and even lower in both the mudstone and the concretion at 54 m, indicating a possible 185 effect of either production of the C_{29} or preferential degradation of the C_{30} hopane during concretion

186 formation. All samples showed high gammacerane ratios (5.6–6.9, Table 1), indicating a stratified water column during deposition in agreement with recent studies of the paleoenvironmental 187 188 conditions during the deposition of the muds of the Gogo Formation (Tulipani et al., 2015b). The 189 ratios of diasteranes over steranes were low across all samples, indicating relatively low thermal 190 maturity (0.12–0.34, Table 1). However, relative amounts of diasteranes were higher in the mudstones 191 than in the concretions. This was expected, as the conversion from steranes to diasteranes is clay-192 catalysed and slower in carbonates than in mudstones (van Kaam-Peters et al., 1998; Nabbefeld et al., 193 2010). Hopane-derived maturity parameters such as C₃₁₋₃₅ 22S/(22S+22R) homohopanes and C_{29} Ts/(C_{29} hopane + C_{29} Ts) did vary between concretions and muds, with the muds showing higher 194 195 amounts of isomerisation of these compounds. Ts/(Tm+Ts), however did not vary as strongly: 196 itranged from 0.30 to 0.40, which is typical for anoxic marine, comparatively immature sediments. 197 Isomerization indices in sulfur-rich sedimentary environments can sometimes vary between facies, 198 and depend on side chain length (Köster et al., 1997). In agreement with this, 22S/(22S+SSR)were, overall, lower for the C₃₁ hopanes (0.4) compared to the C₃₂ to C₃₅ homohopanes (0.6). Concretions 199 200 also showed higher relative amounts of $\beta\beta$ C₃₀ hopane (0.25 and 0.28 in concretions, vs 0.21 and 0.24 in muds). The hopane/sterane ratio for all samples was between 0.17 and 0.19. 201

These results strongly suggest that the biomarker composition of the anoxic mud was preserved during concretion formation. Differences in sterane composition could indicate that part of those are derived from sterols from a former organic matter nucleus. Higher diasterane/sterane , 22S/(22S+22R) hopane, and moretane ratios in the mudstone most likely reflect enhanced isomerisatoin in the presence of clays, which are naturally more abundant in mudstones than in the carbonate concretions.

207 *3.3 Methylhopanes*

Methylhopanes derive from bacterially produced 2- and 3-methylbacteriohopanepolyols and can thus
be useful in the investigation of organic matter sources as well as bacterially mediated processes.
MRM analysis of all samples showed abundant methylhopanes (Supplementary. Fig. 2) which were
present in the muds and the concretions. The 2-methylhopane index (2α-methyl-17α,21β-C₃₁-hopane

/ $(2\alpha$ -methyl-17 α ,21 β -C₃₁-hopane + 17 α ,21 β -C₃₀-hopane) varied between concretions and muds. 212 Concretions are also showing much lower relative amounts of 3β - over 2α -methylhopanes than the 213 muds (0.43 vs. 3.04 and 0.17 vs. 1.12 at 47 m; and 1.28 vs. 2.01 and 0.48 vs. 0.96 at 54 m; for C₃₁ and 214 215 C₃₂ methylhopanes, respectively). While 2-methylhopanes have been considered cyanobacterial biomarkers, it is now clear that a number of different bacteria have the potential to produce their 216 precursors (Welander et al., 2010; Ricci et al., 2014). They are generally considered to be particularly 217 218 prevalent in dysoxic environments (Dumitrescu and Brassell, 2005) and have been reported to be 219 particularly abundant in cyanobacterial akinetes from nutrient-starved Nostoc punctiforme (Doughty 220 et al., 2009). 3-Methylhopanes can be derived from intraaerobic methanotrophs, from acetic acid 221 bacteria, and possibly a range of other sources (Neunlist and Rohmer, 1985; Simonin et al., 1994; Welander and Summons, 2012; Kool et al., 2014) and are generally observed to be elevated in 222 alkaline, lacustrine source rocks (Farrimond et al., 2004). An explanation for the elevated 3-223 224 methylhopanes in the mudstones could be preferential preservation. It is also possible, that the anoxic 225 muds were harbouring microorganisms such as Acetobacter sp. or Methylomirabilis oxyfera (Ettwig 226 et al., 2010), who produce 3-methylhopanoids (Zundel and Rohmer, 1985; Kool et al., 2014) during 227 burial, while the comparatively rapid microbial processes of concretion formation were not involving these microorganisms. This further confirms that the concretion formation occurred very quickly after 228 229 burial.

230 3.4 Principal component analysis

231 Distinctive features confirmed by principal component analysis (PCA) revealed that the first component PC1 could explain differences between mud and concretions (63.10 %), while the second 232 component could explain differences between different depths (23.90 %). Variables that correlated 233 significantly (p < 0.1) with PC1 were partly lithology- and partly source-related ratios (Table 2). The 234 lithology-related indices that showed highest significance in their correlation ($p \le 0.1$) were the 235 236 homohopane isomerisation index C_{31-35} S/(S+R), the diasterane/sterane ratios, some moretane ratios, and the $C_{29}Ts/(C_{29}hopane + C_{29}Ts)$ ratio. These were all higher in the muds, most probably reflecting 237 the higher clay content of the mudstones which catalysed stereochemical conversions. Previously, 238

239 differences in moretane/hopane ratios have been linked to clay contents (French et al., 2012). Other 240 variables that were significantly correlated with PC1, and thus differentiated between concretions and 241 shales were the relative amounts of C₃₀, C₂₉ and C₂₇ steranes and C₃₁/C₃₀ hopanes. As discussed in 242 section 3.3, the higher relative amounts of 3β -methylhopanes in the muds than in the concretions (3β -Me $\alpha\beta$ hopanes/2 α -Me $\alpha\beta$ hopanes) showed strong significance. In turn, variables which 243 differentiated the samples from different depths (PC2) were sterane/hopane ratios, indicating different 244 inputs of organic material or possibly a difference in benthic activity, 2-Methylhopane ratios and % 245 246 C₃₁ homohopanes (Table 2, Supplementary, Fig. 1). Only negligible differences in clay content were present in this core, which is probably the reason for the strong similarities between lithology- or 247 maturity related indicators between the two depths (Table S1). Only further studies including dating 248 249 and paleoenvironmental studies of the core might allow conclusions about the origin of these 250 differences.

251

252 4. Conclusions

Biomarker indices determined from carbonate concretions and their surrounding mudstone in the core 253 254 showed small differences between muds and concretions in those biomarker ratios that are indicative 255 of clay-catalysed conversions in the muds. The stronger differences in source- or process-related 256 biomarker ratios between the muds and the concretions were their methylhopane distributions, which could be due to preferential preservation, or to differences in bacterial processes during concretion 257 formation compared to bacterial activity during burial of the muds; and their C₂₇ sterane abundances, 258 259 which indicates that the concretion formed around labile organic matter such as animal remains. 260 Generally, characteristic biomarker ratios were very similar between concretions and their surrounding muds. This confirms that the organic geochemistry of the carbonates of concretions 261 262 found in a geologically less clear setting such as the eroded Gogo nodules can be used to place them 263 in a paleoenvironmental context.

264

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Table 1. Biomarker ratios for the mudstones and the embedded concretions.

% C_{27} steranes 32 44 37 44 % C_{23} steranes 17 17 15 17 % C_{29} steranes 50 39 47 44 % C_{29} steranes 50 39 47 44 % C_{29} steranes 0.03 0.01 0.02 00 Diasteranes/steranes 0.34 0.12 0.21 0.1 diaC_2/(C_2r+C_{28}+C_{29}) 0.19 0.17 0.17 0.0 diaC_2/(C_{27}+C_{28}+C_{29}) 0.49 0.32 0.46 0.0 % C_{13} homohopanes 18 22 19 14 % C_{14} homohopanes 11 14 13 14 % C_{25} homohopanes 11 14 13 16 C_{34} homohopane 0.28 0.25 0.25 0.0 C_{14} homohopane 0.66 0.58 0.62 0.0 $C_{13} 228/(228+228)$ homohopane 0.66 0.58 0.62 0.0 $C_{13} 228/(228+228)$ homohopane 0.67 0.63 00 0.23 0.23 0.23	Biomarker ratio	Mud 47	Nodule 47	Mud 54	Nodule 54
% C_{23} steranes 17 17 15 14 % C_{23} steranes 50 39 47 4 $C_{10}/(C_{27}-C_{30})$ 0.03 0.01 0.02 0.0 Diasteranes/steranes 0.34 0.12 0.21 0.0 diaC ₂₃ /(C ₂₇ +C ₂₄ +C ₂₉) 0.19 0.17 0.17 0.0 diaC ₂₃ /(C ₂₇ +C ₂₄ +C ₂₉) 0.49 0.32 0.46 0.0 % C ₃₁ homohopanes 36 29 28 23 % C ₃₄ homohopanes 18 22 19 14 % C ₃₄ homohopanes 11 18 15 11 % C ₃₄ homohopanes 11 14 13 15 C_{34} homohopane 0.22 0.16 0.17 0.0 10XGa/(Ga+C ₃₀ hopane 0.56 5.63 5.70 6.88 55 C ₁₄ / ₂₂ Dopane 0.57 0.63 0.62 0.64 00 C ₃₂ 228/(228+22R) homohopane 0.67 0.59 0.65 00 C ₃₂ 228/228/228/228/228/228/228/228/228/22		0.57	0.43	0.47	0.43
% C_{29} stranes 50 39 47 4 $C_{9}/(C_{27}-C_{30})$ 0.03 0.01 0.02 00 Diasteranes/steranes 0.34 0.12 0.21 00 diaC_27(C_{27}+C_{38}+C_{59}) 0.32 0.51 0.37 00 diaC_28/(C_{27}+C_{38}+C_{59}) 0.49 0.32 0.46 00 % C_{31} homohopanes 36 29 28 28 % C_{32} homohopanes 18 22 19 11 % C_{34} homohopanes 11 18 15 11 C_{54} homohopanes 11 14 13 17 C_{50}/C_{30} hopane 0.28 0.25 0.25 0 $C_{1/4}$ bopane 0.28 0.25 0.25 0 0 $C_{1/4}$ bopane 0.66 0.58 0.62 0 0 0 0 0 0 0 22 0.66 0.52 0.64 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 <t< td=""><td>% C₂₇ steranes</td><td>32</td><td>44</td><td>37</td><td>41</td></t<>	% C ₂₇ steranes	32	44	37	41
$\begin{array}{cccc} \zeta_{30}(\overline{C}_{27}-C_{30}) & 0.03 & 0.01 & 0.02 & 0.\\ Diasteranes/steranes & 0.34 & 0.12 & 0.21 & 0.\\ diaC_{27}(C_{27}+C_{28}+C_{29}) & 0.32 & 0.51 & 0.37 & 0.\\ diaC_{29}(C_{27}+C_{28}+C_{29}) & 0.19 & 0.17 & 0.17 & 0.\\ diaC_{29}(C_{27}+C_{28}+C_{29}) & 0.49 & 0.32 & 0.46 & 0.\\ \% C_{31} homohopanes & 36 & 29 & 28 & 22 & 0.\\ \% C_{31} homohopanes & 18 & 22 & 19 & 19 & 0.\\ \% C_{32} homohopanes & 18 & 22 & 19 & 19 & 0.\\ \% C_{34} homohopanes & 11 & 18 & 15 & 17 & 0.\\ \% C_{35} homohopanes & 11 & 14 & 13 & 17 & 0.\\ \% C_{35} homohopanes & 11 & 14 & 13 & 17 & 0.\\ \zeta_{97}(C_{10} hopane & 0.22 & 0.16 & 0.17 & 0.\\ C_{31}/C_{30} hopane & 0.22 & 0.16 & 0.17 & 0.\\ C_{31}/C_{30} hopane & 0.22 & 0.16 & 0.17 & 0.\\ OC_{31} 2S/(228+22R) homohopane & 0.66 & 0.58 & 0.62 & 0.\\ C_{31} 2S/(228+22R) homohopane & 0.69 & 0.62 & 0.64 & 0.\\ C_{32} 22S/(228+22R) homohopane & 0.67 & 0.60 & 0.64 & 0.\\ C_{32} 22S/(228+22R) homohopane & 0.67 & 0.60 & 0.64 & 0.\\ C_{32} 22S/(228+22R) homohopane & 0.67 & 0.59 & 0.65 & 0.\\ C_{31} C_{31}(C_{20} hopane + C_{20} T_8) & 0.27 & 0.14 & 0.25 & 0.\\ D/(\alpha\beta+\beta\alpha+\beta\beta+\alpha+\beta\beta) C_{31} hopane & 0.05 & 0.05 & 0.04 & 0.\\ D/(\alpha\beta+\beta\alpha+\beta\beta+\alpha+\beta\beta) C_{31} hopane & 0.05 & 0.05 & 0.04 & 0.\\ D/(\alpha\beta+\beta\alpha+\beta\beta+\alpha+\beta\beta) C_{32} hopane & 0.19 & 0.18 & 0.17 & 0.\\ D/(\alpha\beta+\beta\alpha+\beta\beta) C_{32} hopane & 0.19 & 0.18 & 0.17 & 0.\\ D/(\alpha\beta+\beta\alpha+\beta\alpha+2\alpha-Me \alpha\beta C_{31} hopane & 1.59 & 1.02 & 1.75 & 1.\\ 2\beta-Me \alpha\beta C_{31} hopane/2\alpha-Me \alpha\beta C_{31} hopane & 1.59 & 1.02 & 1.75 & 1.\\ 2\beta-Me \alpha\beta C_{31} hopane/2\alpha-Me \alpha\beta C_{31} hopane & 1.16 & 0.96 & 1.54 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane/2\alpha-Me \alpha\beta C_{31} hopane & 0.15 & 0.18 & 0.38 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{32} hopane & 0.22 & 0.21 & 0.25 & 0.24 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.26 & 0.13 & 0.22 & 0.28 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.15 & 0.18 & 0.38 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.26 & 0.13 & 0.25 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.26 & 0.13 & 0.25 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.32 & 0.22 & 0.28 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.26 & 0.13 & 0.25 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.26 & 0.13 & 0.99 & 0.12 & 0.\\ D/(\alpha\beta+\beta\alpha) C_{31} hopane & 0.$	% C ₂₈ steranes	17	17	15	15
$\begin{array}{llllllllllllllllllllllllllllllllllll$	% C ₂₉ steranes	50	39	47	44
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$C_{30}/(C_{27}-C_{30})$	0.03	0.01	0.02	0.02
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Diasteranes/steranes	0.34	0.12	0.21	0.19
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$diaC_{27}/(C_{27}+C_{28}+C_{29})$	0.32	0.51	0.37	0.41
% C_{31} homohopanes 36 29 28 28 % C_{32} homohopanes 18 22 19 19 % C_{33} homohopanes 24 17 25 22 % C_{34} homohopanes 11 18 15 11 % C_{30} homohopanes 11 14 13 11 C_{25}/C_{30} hopane 0.22 0.16 0.17 00 10xGa/(Ga+C_{30} hopane) 5.63 5.70 6.88 5.5 $C_{3,1,52}22S/(22S+22R)$ homohopane 0.66 0.58 0.62 0.64 $C_{32}22S/(22S+22R)$ homohopane 0.67 0.63 0.64 00 $C_{32}22S/(22S+22R)$ homohopane 0.67 0.59 0.65 00 $C_{32}2S/(22S+22R)$ homohopane 0.67 0.59 0.65 00 $C_{32}(2S+22R)$ homohopane 0.27 0.14	$diaC_{28}/(C_{27}+C_{28}+C_{29})$	0.19	0.17	0.17	0.16
% C_{32} homohopanes 18 22 19 11 % C_{33} homohopanes 11 18 15 1 % C_{34} homohopanes 11 18 15 1 % C_{34} homohopanes 11 14 13 1 C_{39}/C_{30} hopane 0.22 0.16 0.17 00 C_{31/C_{30} hopane 0.22 0.16 0.17 00 C_{31/C_{30} hopane 0.22 0.16 0.17 00 $C_{31/C_{32}}(22S+22R)$ homohopane 0.66 0.58 0.62 0.64 00 $C_{32}(22S/22S+22R)$ homohopane 0.67 0.60 0.64 00 02 0.53 22S/(22S+22R) homohopane 0.67 0.60 0.64 00 02 0.53 22S/(22S+22R) homohopane 0.67 0.60 0.64 00 02 0.53 22S/(22S+22R) homohopane 0.67 0.50 0.65 00 0.5 0.27 0.14 0.25 0.27 0.14 0.25 0.27 0.14 0.25 0.27 0.14 0.25 0.27 0.14 0.25 0.27 <	$diaC_{29}/(C_{27}+C_{28}+C_{29})$	0.49	0.32	0.46	0.43
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	$% C_{31}$ homohopanes	36	29	28	36
% C_{34} homohopanes 11 18 15 1 % C_{35} homohopanes 11 14 13 1 C_{29}/C_{30} hopane 0.28 0.25 0.25 0 C_{31}/C_{30} hopane 0.22 0.16 0.17 0 10xGa/(Ga+C_{30} hopane) 5.63 5.70 6.88 5.5 C_{31-35} 22S/(22S+22R) homohopane 0.66 0.58 0.62 0.64 C_{31} 22S/(22S+22R) homohopane 0.68 0.57 0.63 00 C_{32} 22S/(22S+22R) homohopane 0.67 0.60 0.64 0 C_{32} 22S/(22S+22R) homohopane 0.67 0.59 0.65 0 C_{32} 22S/(22S+22R) homohopane 0.67 0.59 0.65 0 C_{32} 22S/(22S+22R) homohopane 0.67 0.59 0.65 0 C_{25} 22S/(22S+22R) homohopane 0.67 0.59 0.65 0 C_{25} 22S/(22S+22R) homohopane 0.67 0.59 0.65 0 C_{25} 2S/(22S+22R) homohopane 0.67 0.59 0.65 0 0 C_{32} Dipane/2 α -M	$% C_{32}$ homohopanes	18	22	19	19
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	% C ₃₃ homohopanes	24	17	25	21
% C_{35} homohopanes 11 14 13 14 C_{2y}/C_{30} hopane 0.28 0.25 0.25 0.0 C_{31}/C_{30} hopane 0.22 0.16 0.17 0.0 $10xGa/(Ga+C_{30}$ hopane) 5.63 5.70 6.88 5.70 C_{31-35} 22S/(22S+22R) homohopane 0.66 0.58 0.62 0.64 C_{31} 22S/(22S+22R) homohopane 0.69 0.62 0.64 0.0 C_{32} 22S/(22S+22R) homohopane 0.67 0.60 0.64 0.0 C_{33} 22S/(22S+22R) homohopane 0.67 0.60 0.64 0.0 C_{32} 22S/(22S+22R) homohopane 0.67 0.59 0.65 0.0 C_{33} 22S/(22S+22R) homohopane 0.67 0.59 0.65 0.0 C_{37} 52S/(22S+22R) homohopane 0.67 0.59 0.65 0.0 C_{37} 52S/(22S+22R) homohopane 0.67 0.59 0.65 0.0 C_{37} 52S/(22S+22R) homohopane 0.67 0.59 0.65 0.0 C_{37} Tor That 0.25 0.24 0.0 0.66 0.03 0.06	-	11	18	15	13
$\begin{array}{cccc} C_{31}/C_{30} \ \ \ hopane & 0.22 & 0.16 & 0.17 & 0. \\ 10xGa/(Ga+C_{30} \ \ hopane) & 5.63 & 5.70 & 6.88 & 5. \\ C_{31-35} 22S/(22S+22R) \ \ \ homohopane & 0.66 & 0.58 & 0.62 & 0. \\ C_{31} 22S/(22S+22R) \ \ \ \ homohopane & 0.69 & 0.62 & 0.64 & 0. \\ C_{32} 22S/(22S+22R) \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	$% C_{35}$ homohopanes	11	14	13	11
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	· · · · · · · · · · · · · · · · · · ·	0.28	0.25	0.25	0.21
$\begin{array}{llllllllllllllllllllllllllllllllllll$	· · · · ·	0.22	0.16	0.17	0.16
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-		5.70	6.88	5.82
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C3222S/(22S+22R) homohopane0.690.620.640.0C3322S/(22S+22R) homohopane0.680.570.630.0C3422S/(22S+22R) homohopane0.670.600.640.0C3522S/(22S+22R) homohopane0.670.590.650.0C3522S/(22S+22R) homohopane0.670.590.650.0C3522S/(22S+22R) homohopane0.670.590.650.0C3522S/(22S+22R) homohopane0.670.590.650.0C3522S/(22S+22R) homohopane0.670.590.650.0C3522S/(22S+22R) homohopane0.670.590.650.0C3522S/(22S+22R) homohopane0.670.590.650.0C3522S/(22S+22R) homohopane0.320.400.360.0C3522S/(22S+22R) homohopane0.210.250.240.0B/(\abla\beta\beta\beta\beta\beta\beta\beta\bet					0.44
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					0.62
$\begin{array}{ccccc} C_{34} 22S/(22S+22R) \ \text{homohopane} & 0.67 & 0.60 & 0.64 & 0. \\ C_{35} 22S/(22S+22R) \ \text{homohopane} & 0.67 & 0.59 & 0.65 & 0. \\ C_{29} Ts/(C_{29} \ \text{hopane} + C_{29} Ts) & 0.27 & 0.14 & 0.25 & 0. \\ Ts/(Tm+Ts) & 0.32 & 0.40 & 0.36 & 0. \\ \beta\beta/(\alpha\beta+\beta\alpha+\beta\beta) C_{30} \ \text{hopane} & 0.21 & 0.25 & 0.24 & 0. \\ \beta\beta/(\alpha\beta+\beta\alpha+\beta\beta) C_{31} \ \text{hopane} & 0.05 & 0.05 & 0.04 & 0. \\ \beta\beta/(\alpha\beta+\beta\alpha+\beta\beta) C_{32} \ \text{hopane} & 0.06 & 0.03 & 0.06 & 0. \\ Hopanes / steranes & 0.19 & 0.18 & 0.17 & 0. \\ 2-Methylhopane \ \text{index} & 0.12 & 0.30 & 0.39 & 0. \\ 2\beta-Me \ \alpha\beta C_{31} \ \text{hopane}/2\alpha-Me \ \alpha\beta C_{31} \ \text{hopane} & 1.59 & 1.02 & 1.75 & 1. \\ 2\beta-Me \ \alpha\beta C_{32} \ \text{hopane}/2\alpha-Me \ \alpha\beta C_{31} \ \text{hopane} & 1.16 & 0.96 & 1.54 & 0. \\ 3\beta-Me \ \alpha\beta C_{31} \ \text{hopane}/2\alpha-Me \ \alpha\beta C_{31} \ \text{hopane} & 3.04 & 0.43 & 2.01 & 1. \\ 3\beta-Me \ \alpha\beta C_{32} \ \text{hopane}/2\alpha-Me \ \alpha\beta C_{32} \ \text{hopane} & 1.12 & 0.17 & 0.96 & 0. \\ Moretane / \ hopane \ ratios & Moretane / \ hopane \ ratios & Moretane / \ hopane \ ratios & Moretane / \ hopane & 0.26 & 0.13 & 0.25 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ 2\alpha-Me \ C_{31} \ \text{hopane} & 0.26 & 0.13 & 0.25 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{30} \ \text{hopane} & 0.13 & 0.09 & 0.12 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.13 & 0.09 & 0.12 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.20 & 0.13 & 0.19 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.25 & 0.18 & 0.21 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{31} \ \text{hopane} & 0.25 & 0.18 & 0.21 & 0. \\ \beta\alpha/(\alpha\beta + \beta\alpha) \ C_{$					0.61
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					0.62
$C_{29}Ts/(C_{29} hopane + C_{29} Ts)$ 0.27 0.14 0.25 0.014 $Ts/(Tm+Ts)$ 0.32 0.40 0.36 0.015 $\beta\beta/(\alpha\beta+\beta\alpha+\beta\beta) C_{30} hopane$ 0.21 0.25 0.24 0.016 $\beta\beta/(\alpha\beta+\beta\alpha+\beta\beta) C_{31} hopane$ 0.05 0.05 0.04 0.016 $\beta\beta/(\alpha\beta+\beta\alpha+\beta\beta) C_{32} hopane$ 0.06 0.03 0.06 0.016 $hopanes / steranes$ 0.19 0.18 0.17 0.12 $2-Methylhopane index$ 0.12 0.30 0.39 0.26 $2-Methylhopane index$ 0.12 0.30 0.39 0.26 $2\beta-Me \alpha\beta C_{31} hopane/2\alpha-Me \alpha\beta C_{32} hopane1.530.941.981.262\beta-Me \alpha\beta C_{32} hopane/2\alpha-Me \alpha\beta C_{32} hopane1.160.961.540.663\beta-Me \alpha\beta C_{31} hopane/2\alpha-Me \alpha\beta C_{32} hopane1.120.170.960.16\beta\alpha/(\alpha\beta + \beta\alpha) 2\alpha-Me C_{32} hopane0.150.180.380.66\beta\alpha/(\alpha\beta + \beta\alpha) 2\alpha-Me C_{32} hopane0.290.130.240.66\beta\alpha/(\alpha\beta + \beta\alpha) 2\alpha-Me C_{32} hopane0.260.130.250.26\beta\alpha/(\alpha\beta + \beta\alpha) C_{29} hopane0.320.220.280.66\beta\alpha/(\alpha\beta + \beta\alpha) C_{31} hopane0.130.090.120.66/(\alpha\beta + \beta\alpha) C_{32} hopane\beta\alpha/(\alpha\beta + \beta\alpha) C_{31} hopane0.200.130.190.16\beta\alpha/(\alpha\beta + \beta\alpha) C_{31} hopane0.250.180.210.66/(\alpha\beta + \beta\alpha) C_{32} hopane\beta\alpha/(\alpha\beta + \beta\alpha) C_{31} hopane<$	· · · · · -				0.64
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$\begin{array}{cccccccccccccccccccccccccccccccccccc$					0.30
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					0.28
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Hopmes / steranes0.190.180.170.22-Methylhopane index0.120.300.390.22-Methylhopane index0.120.300.390.22β-Me αβ C ₃₁ hopane/2α-Me αβ C ₃₁ hopane1.591.021.751.22β-Me αβ C ₃₂ hopane/2α-Me αβ C ₃₂ hopane1.530.941.981.22β-Me αβ C ₃₃ hopane/2α-Me αβ C ₃₁ hopane1.160.961.540.23β-Me αβ C ₃₁ hopane/2α-Me αβ C ₃₁ hopane3.040.432.011.23β-Me αβ C ₃₂ hopane/2α-Me αβ C ₃₂ hopane1.120.170.960.2Moretane / hopane ratios0.150.180.380.240.2βα/(αβ + βα) 2α-Me C ₃₂ hopane0.260.130.250.2βα/(αβ + βα) 2α-Me C ₃₁ hopane0.320.220.280.2βα/(αβ + βα) C ₂₉ hopane0.130.090.120.2βα/(αβ + βα) C ₃₀ hopane0.130.090.120.2βα/(αβ + βα) C ₃₁ hopane0.200.130.190.2βα/(αβ + βα) C ₃₂ hopane0.200.130.190.2βα/(αβ + βα) C ₃₁ hopane0.200.130.190.2βα/(αβ + βα) C ₃₂ hopane0.250.180.210.2βα/(αβ + βα) C ₃₃ hopane0.250.180.210.2					0.05
2-Methylhopane index0.120.300.390. 2β -Me $\alpha\beta$ C ₃₁ hopane/2 α -Me $\alpha\beta$ C ₃₁ hopane1.591.021.751. 2β -Me $\alpha\beta$ C ₃₂ hopane/2 α -Me $\alpha\beta$ C ₃₂ hopane1.530.941.981. 2β -Me $\alpha\beta$ C ₃₃ hopane/2 α -Me $\alpha\beta$ C ₃₃ hopane1.160.961.540. 2β -Me $\alpha\beta$ C ₃₁ hopane/2 α -Me $\alpha\beta$ C ₃₁ hopane3.040.432.011. 3β -Me $\alpha\beta$ C ₃₂ hopane/2 α -Me $\alpha\beta$ C ₃₂ hopane1.120.170.960. 3β -Me $\alpha\beta$ C ₃₂ hopane/2 α -Me $\alpha\beta$ C ₃₂ hopane0.150.180.380. $\beta\alpha/(\alpha\beta + \beta\alpha)$ 2 α -Me C ₃₃ hopane0.290.130.240. $\beta\alpha/(\alpha\beta + \beta\alpha)$ 2 α -Me C ₃₁ hopane0.260.130.250. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₂₉ hopane0.320.220.280. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₃₀ hopane0.130.090.120. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₃₁ hopane0.200.130.190. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₃₂ hopane0.220.280. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₃₁ hopane0.200.130.190. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₃₁ hopane0.200.130.190. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₃₂ hopane0.180.090.140. $\beta\alpha/(\alpha\beta + \beta\alpha)$ C ₃₁ hopane0.250.180.210.					0.19
2β-Me αβ C31 hopane/2α-Me αβ C31 hopane1.591.021.751.752β-Me αβ C32 hopane/2α-Me αβ C32 hopane1.530.941.981.752β-Me αβ C33 hopane/2α-Me αβ C33 hopane1.160.961.540.753β-Me αβ C31 hopane/2α-Me αβ C31 hopane3.040.432.011.753β-Me αβ C32 hopane/2α-Me αβ C32 hopane1.120.170.960.753β-Me αβ C32 hopane/2α-Me αβ C32 hopane1.120.170.960.75βα/(αβ + βα) 2α-Me C33 hopane0.150.180.380.76βα/(αβ + βα) 2α-Me C32 hopane0.260.130.250.75βα/(αβ + βα) 2α-Me C31 hopane0.320.220.280.76βα/(αβ + βα) C29 hopane0.130.090.120.76βα/(αβ + βα) C30 hopane0.130.090.120.76βα/(αβ + βα) C31 hopane0.200.130.190.76βα/(αβ + βα) C31 hopane0.200.130.190.76βα/(αβ + βα) C31 hopane0.250.180.210.76	-				0.15
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					1.25
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
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$\begin{array}{cccccccccccccccccccccccccccccccccccc$					0.89
Moretane / hopane ratios $\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C33 hopane0.150.180.380.6 $\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C32 hopane0.290.130.240.6 $\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C31 hopane0.260.130.250.6 $\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C31 hopane0.320.220.280.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{29}$ hopane0.130.090.120.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{30}$ hopane0.130.090.120.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane0.180.090.140.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.180.090.140.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane0.250.180.210.6					1.28
$\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C33 hopane0.150.180.380.6 $\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C32 hopane0.290.130.240.6 $\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C31 hopane0.260.130.250.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{29}$ hopane0.320.220.280.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{30}$ hopane0.130.090.120.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{30}$ hopane0.200.130.190.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane0.180.090.140.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.180.090.140.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane0.250.180.210.6		1.12	0.17	0.96	0.48
$\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C32 hopane0.290.130.240.9 $\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C31 hopane0.260.130.250.9 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{29}$ hopane0.320.220.280.9 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{30}$ hopane0.130.090.120.9 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{30}$ hopane0.200.130.190.9 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane0.180.090.140.9 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.180.090.140.9 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane0.250.180.210.9	-				
$\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C31 hopane0.260.130.250.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{29}$ hopane0.320.220.280.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{30}$ hopane0.130.090.120.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane0.200.130.190.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.180.090.140.6 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.250.180.210.6					0.19
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$\beta \alpha / (\alpha \beta + \beta \alpha) C_{30}$ hopane 0.13 0.09 0.12 0.30 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane 0.20 0.13 0.19 0.13 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane 0.18 0.09 0.14 0.13 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane 0.25 0.18 0.21 0.21	$3\alpha/(\alpha\beta + \beta\alpha) 2\alpha$ -Me C ₃₁ hopane	0.26	0.13	0.25	0.17
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane0.200.130.190.13 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.180.090.140.13 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane0.250.180.210.13	$3\alpha/(\alpha\beta + \beta\alpha) C_{29}$ hopane	0.32	0.22	0.28	0.24
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane0.200.130.190.13 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.180.090.140.13 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane0.250.180.210.13	$\frac{\beta\alpha}{(\alpha\beta + \beta\alpha)} C_{30}$ hopane	0.13	0.09	0.12	0.07
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane0.180.090.140.9 $\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane0.250.180.210.9		0.20	0.13	0.19	0.15
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane 0.25 0.18 0.21 0.					0.14
					0.20
$\mathbf{p}_{\mathbf{n}}(\mathbf{n}\mathbf{p} + \mathbf{p}_{\mathbf{n}}) = \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} = \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} = \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} = \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} = \mathbf{g}_{\mathbf{n}} = \mathbf{g}_{\mathbf{n}} + \mathbf{g}_{\mathbf{n}} = \mathbf{g}$					0.18
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{35}$ hopane 0.20 0.14 0.18 0.					0.15

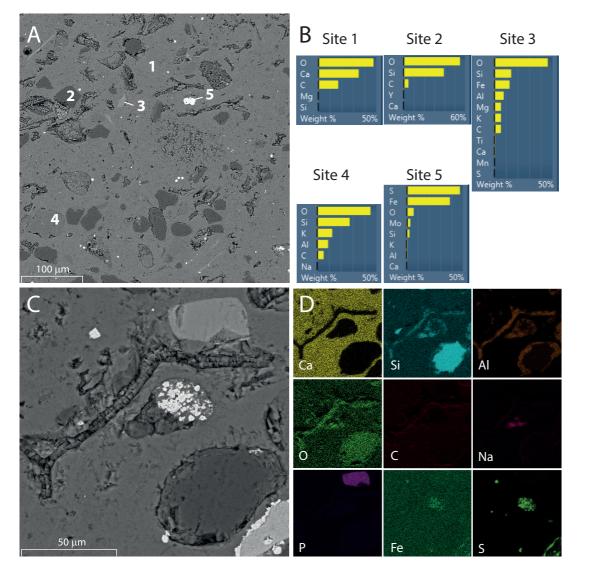
- 443 Table 2. Correlation coefficients with p-values for the ratios determining the first (PC1, 63.10 % of variability)
- 444 and second component (PC2, 23.90 % of variability).

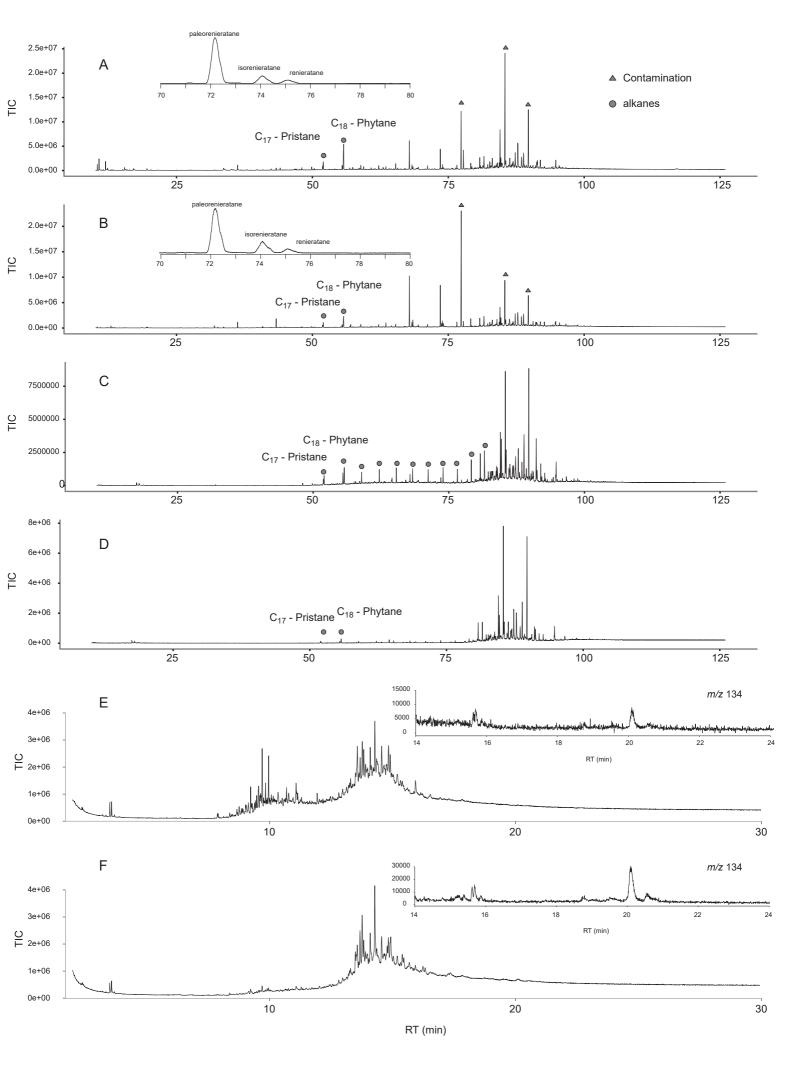
Biomarker ratio	Correlation	p-value
PC1 (Mud vs Nodule	e)	
3β-Me $\alpha\beta$ C_{31} hopane/2α-Me $\alpha\beta$ C_{31} hopane	0.997	0.003
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{34}$ hopane	0.988	0.012
% C ₂₉ steranes	0.988	0.012
3β-Me $\alpha\beta$ C_{32} hopane/2α-Me $\alpha\beta$ C_{32} hopane	0.986	0.014
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{35}$ hopane	0.981	0.019
C31-35 22S/(22S+22R) homohopanes	0.980	0.020
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{29}$ hopane	0.977	0.023
C_{29} Ts/(C_{29} hopane + C_{29} Ts)	0.975	0.025
$\beta \alpha / (\alpha \beta + \beta \alpha) 2 \alpha$ -Me C ₃₁ hopane	0.975	0.025
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{31}$ hopane	0.971	0.029
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{33}$ hopane	0.955	0.045
Diasteranes/Steranes	0.952	0.048
$\beta \alpha / (\alpha \beta + \beta \alpha) C_{32}$ hopane	0.940	0.060
% C ₃₃ homohopanes	0.921	0.079
$\beta\beta/(\alpha\beta+\beta\alpha+\beta\beta) C_{32}$ hopane	0.918	0.082
$C_{30}/(C_{27}-C_{30})$	0.916	0.084
% C ₃₂ homohopanes	-0.928	0.072
% C ₂₇ steranes	-0.992	0.008

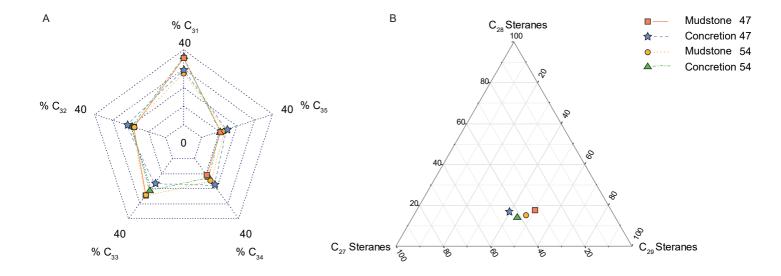
PC2 (47 vs 54 m)		
2-Methylhopane index	0.953	0.047
% C ₃₁ homohopanes	-0.948	0.052
Hopanes / steranes	-0.982	0.018

447 Figure captions.

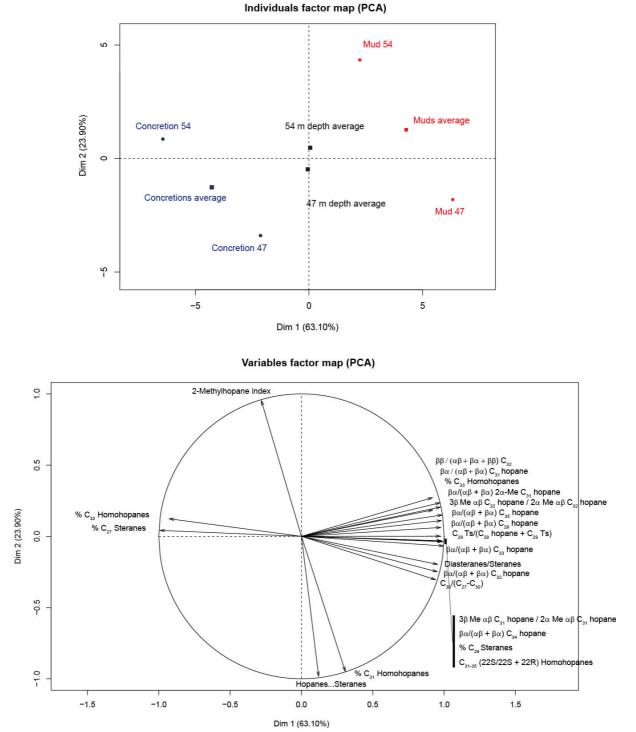
- 448 Figure 1. Scanning electron microscopy images of one concretion. A shows a backscatter image, and B shows
- the elemental mapping results for the different types of mineral seen in the backscatter image. C shows a smaller
- 450 section backscatter image, and D shows element abundances / K of the section shown in C.
- 451 Figure 2. Chromatograms of TLEs and fractions. A BSTFA-derivatised TLE of the concretion at 47 m, B
- 452 BSTFA-derivatised TLE of the concretion at 54 m. C Saturates fraction of the mudstone at 47 m, D Saturates
- 453 fraction of the mudstone at 54 m, E Aromatic fraction of the mudstone at 47 m, F aromatic fraction of the
- 454 mudstone at 54 m. Inserts in E and F show the extracted ion current at m/z 134, while inserts in A and B show
- 455 the GC-MRM results of the TLEs for transitions m/z 546 \rightarrow 134 and the peaks identified by comparison to
- 456 Blina-oil standard.
- 457 **Figure 3.** Biomarker ratios of the mudstones and concretions. Shown are % of homohopanes (A) and ternary
- 458 diagram of sterane distributions (B).



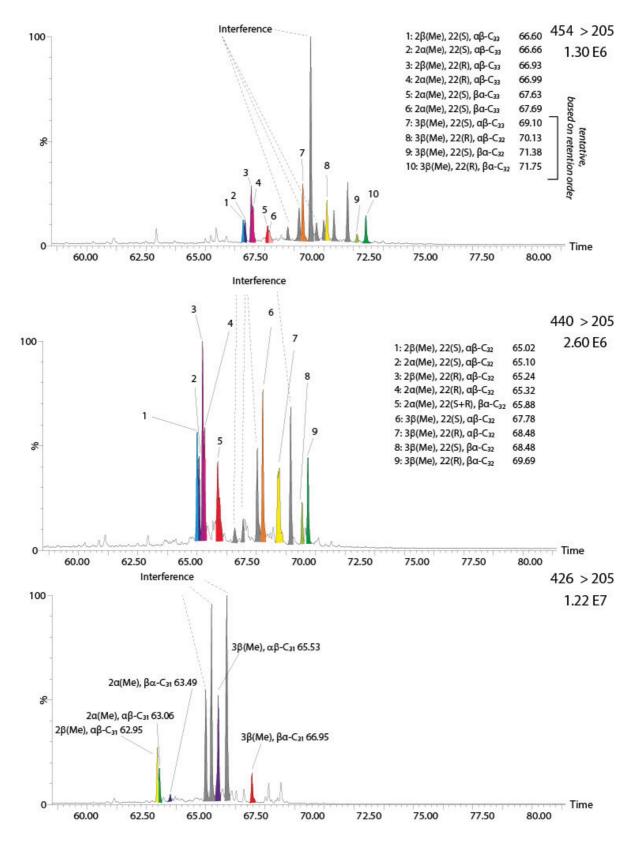




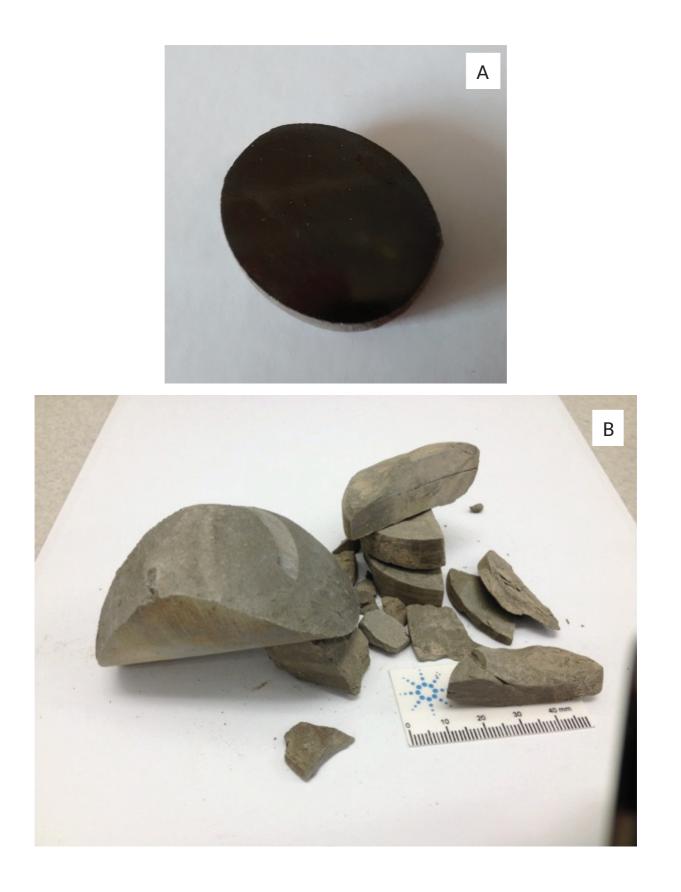
Supplementary Material to *Mudstones and embedded concretions show differences in lithologyrelated, but not source-related biomarker distributions* by Sabine K. Lengger, Ines M. Melendez, and Kliti Grice.



Supplementary Figure 1. Results of the principal component analysis (only significant factors included in the graph).



Supplementary Figure 2. Identified methylhopanes.



Supplementary Figure 3. Pictures of the mud and concretion. (A) 47 m depth (concretion), (B) 54 m depth (mudstones and concretion).

Supplementary Table 1. QXRD analysis results

Quantitative analysis [% w/w]		
Mudstone 47.15 m	Mudstone 54.46 m	
4.6	6.9	
1.8	3.0	
2.4	2.1	
9.5	10	
22	27	
6.4	6.7	
53	44	
	Mudstone 47.15 m 4.6 1.8 2.4 9.5 22 6.4	