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# Effects of Ocean Warming and Acidification on Rhodolith/Maerl Beds

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#### 6 Chapter 4: Effects of ocean warming and acidification on rhodolith/maerl beds

#### 7 S. MARTIN & J. M. HALL-SPENCER

8 Ocean warming and acidification are currently under scrutiny since in combination they are

- 9 expected to reduce marine biodiversity and profoundly alter ecosystem function. Habitats
- 10 where coralline algae play an important ecological role, such as rhodolith and maerl beds, are
- 11 threatened by dissolution due to the high solubility of their high-magnesium calcite skeletons.
- 12 Those in the Arctic, where carbonate saturation levels are already low and falling, are of
- particular concern. As well as direct corrosive effects on coralline algae, rising CO<sub>2</sub> emissions
   are expected to have knock-on effects on ecosystems because reduced seabed habitat
- 15 complexity causes a reduction in biodiversity and simplifies food webs. We anticipate that
- 16 degradation of coastal calcareous habitats due to ocean acidification will facilitate a
- 17 proliferation of fleshy algae that may benefit from an increase in dissolved inorganic carbon,
- to the detriment of calcified algae. It is not all doom and gloom: coralline algae have survived
- 19 previous mass extinctions and many species tolerate highly variable CO<sub>2</sub> levels. Which
- 20 species survive the Anthropocene will depend upon their ability to acclimate and adapt. These
- 21 topics warrant further research since coralline algae provide highly biodiverse habitats that
- 22 benefit commercially important species of fish and molluscs. Data so far suggest that 1) this
- 23 important algal group is especially vulnerable to ocean acidification and warming and 2)
- 24 protecting these habitats has long-term benefits, not least because coralline algae provide
- 25 habitat for species of commercial importance.
- 26

## 27 4.1 Climate change and ocean acidification

28 Anthropogenic emissions have increased the atmospheric carbon dioxide (CO<sub>2</sub>) concentration

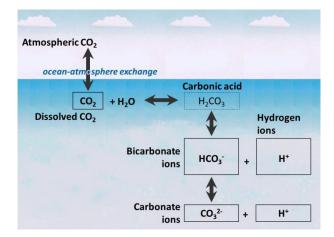
from 280 ppm prior to the beginning of the industrial revolution (1750) to more than 390 ppm

30 in 2013; this is *ca*. 100 ppm higher than at any time in the past 740,000 years. The present rate

31 of increase in atmospheric CO<sub>2</sub> and temperature is unprecedented in recent Earth history

32 (Hoegh-Guldberg et al. 2007).

- 33 Approximately 25 % of CO<sub>2</sub> emissions dissolve into the ocean, increasing pCO<sub>2</sub> in the surface
- 34 water and altering seawater carbonate chemistry (Canadell et al. 2007). This CO<sub>2</sub> reacts with
- 35 the water to form carbonic acid most of which dissociates into ions of hydrogen (H<sup>+</sup>) and
- 36 bicarbonate ( $CO_3^{2-}$ ) (Fig. 1). The increased concentration of H<sup>+</sup> reduces pH (pH =  $-log_{10}[H^+]$ )
- 37 and carbonate ion  $(CO_3^{2-})$  concentration and increases the concentration of  $HCO_3^{-}$ .

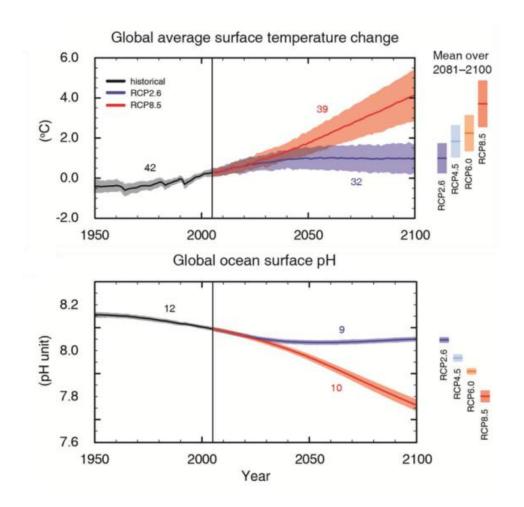


- 39 Fig. 1. Equilibrium between the three different species of dissolved inorganic carbon in seawater.
- 40 Atmospheric CO<sub>2</sub> is absorbed at the surface of the oceans where it combines with the water molecules to 41 form carbonic acid, which dissociates into bicarbonate, carbonate, and hydrogen ions. Different forms of
- 41 form carbonic acid, which dissociates into bicarbonate, carbonate, and hydrogen ions. Different forms of 42 dissolved inorganic carbon (DIC =  $[CO_2] + [HCO_3^-] + [CO_3^2^-]$ ) follow thermodynamic equilibria in
- 43 seawater for the following reactions:  $CO_2 + H_2O \leftrightarrow H_2CO_3 \leftrightarrow HCO_3^- \leftrightarrow H^+ + CO_3^{2-} + 2H^+$ .

44

- 45 During the  $20^{\text{th}}$  century, increasing atmospheric CO<sub>2</sub> caused the surface ocean to warm by
- 46 0.7°C and to acidify by 0.1 pH units (Hoegh-Guldberg et al. 2007). The atmospheric CO<sub>2</sub>
- 47 concentration is expected to reach between 490 1370 ppm by 2100 causing a global mean
- 48 surface temperature increase of between 0.3 to 4.8°C and a surface ocean pH decrease of
- 49 0.06-0.32 units relative to the period 1986-2005 (IPCC 2013; Fig. 2). Ocean acidification and
- 50 warming will not be regionally uniform; marine organisms in the Arctic region are thought to
- 51 face more rapid and stronger warming and acidification than the global mean (Fig. 3).

52

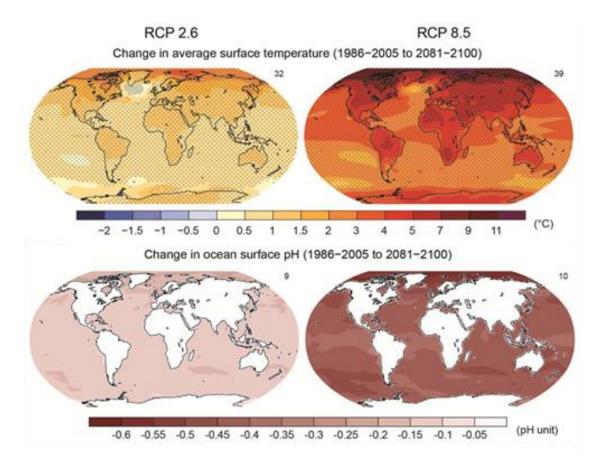


53

Fig. 2. Projected change in global annual mean surface temperature and global mean ocean surface pH,

54 55 56 relative to 1986-2005 for Representative Concentration Pathway 2.6 (blue) and RCP8.5 (red) (Source: IPCC 2013).

57



58

Fig. 3. Annual mean temperature and pH changes for the scenarios RCP2.6 and RCP8.5 in 2081-2100
 relative to 1986-2005 (Source: IPCC 2013).

#### 62 **4.2 Effects of ocean warming and acidification on seaweeds**

63 Global warming and ocean acidification will alter biochemical and physiological processes in seaweeds, causing changes in their ecological interactions (Koch et al. 2012, Harley et al. 64 65 2012). Here we base predictions about the fate of rhodolith/maerl beds upon a growing body of information about the effects of warming and acidification on seaweeds in general and on 66 coralline algae in particular. While some seaweed taxa seem likely to benefit from ocean 67 warming and acidification, coralline algae are expected to be adversely impacted, putting the 68 habitats they form at risk. This is a concern since beds of rhodoliths and maerl form highly 69 70 biodiverse but very slow-growing habitats (Pena et al. 2014).

71

#### 72 **4.2.1 Effects of ocean warming on seaweeds**

73 Temperature directly affects enzymatic processes and so alters rates of photosynthesis and

74 growth; it also determines whether seaweeds can survive and reproduce and sets limits to their

- 75 geographic distribution (Lüning 1990). Seaweeds, as with other organisms, have thermal
- 76 performance curves that show beneficial effects of increased temperature at suboptimal
- temperature but detrimental effects above a threshold (Lüning 1990). In order for some
- 78 individual species of seaweed to survive rising sea surface temperature they will need to
- 79 acclimate (on short timescales), adapt (on medium and long timescales) or migrate (by

dispersion) (Wiencke & Bischof 2012). Seaweeds that are intertidal or have a broad

- 81 biogeographic range are more tolerant of temperature extremes than subtidal species and
- those with narrower temperature ranges. The changes in temperature we see today are driving
   a poleward shift in seaweed biogeographical regions (Wernberg et al. 2011). The tropical
- region is widening polewards, to the detriment of organisms that occupy warm-temperate
- regions and the cold-temperate regions are shrinking. Arctic seaweeds are at particular risk
- 86 due to warming winter temperatures (Wiencke & Bischof 2012). In coralline algae, a small
- 87 rise in temperature, within the range of temperature experienced in natural habitats, can
- 88 increase growth, photosynthesis and calcification in both temperate and tropical species
- 89 (Martin et al. 2006, Steller et al. 2007) but this has limits and rising temperature above these
- levels is detrimental (Table 1). For example, Agegian (1985) showed that growth of tropical
   *Porolithon gardineri* slowed dramatically at temperatures above 29-30°C. An increase of
- $+3^{\circ}$ C above that normally experienced by coralline algae causes bleaching and adversely
- 93 affects health, survival, and the rates of photosynthesis and calcification in both tropical and
- 94 warm-temperate coralline algae (Anthony et al. 2008, Martin & Gattuso 2009, Diaz-Pulido et
- 95 al. 2011, Martin et al. 2013; Table 1).
- 96

# 97 **4.2.2 Impact of ocean acidification on seaweeds**

98 Decreasing pH in the surface ocean will cause major shifts in seawater chemistry over the 99 course of this century that are likely to affect photosynthesis and calcification since these 100 processes use dissolved inorganic carbon (DIC:  $HCO_3^-$ ,  $CO_3^{2-}$  and  $CO_2$ ) as substrate

- 101 (1) Photosynthesis (left to right) and respiration (right to left) processes:
- 102  $CO_2 + H_2O \leftrightarrow CH_2O + O_2$

103

 $HCO_3^- + H_2O \leftrightarrow CH_2O + O_2 + OH^-$ 

- 104 (2) Gross calcification (left to right) and dissolution (right to left) processes:
- 105  $CO_3^{2-} + Ca^{2+} \leftrightarrow CaCO_3$

106 
$$2 \operatorname{HCO}_3^- + \operatorname{Ca}^{2+} \leftrightarrow \operatorname{CaCO}_3 + \operatorname{H}_2\operatorname{O} + \operatorname{CO}_2$$

107 Algae can use dissolved CO<sub>2</sub> entering the cell by diffusion as the carbon source for 108 photosynthesis but most of them have carbon concentrating mechanisms which actively take 109 up  $HCO_3^-$  which is then converted to  $CO_2$  in the cells (Raven et al. 2002, Hepburn et al. 2011, 110 Cornwall et al. 2012). This mechanism counteracts the limited availability of CO<sub>2</sub> in seawater 111 (Raven & Beardall 2003). An increase in seawater  $pCO_2$  is expected to enhance photosynthesis 112 in primary producers that rely exclusively on CO<sub>2</sub> diffusion (Kübler et al. 1991) and may be 113 favourable to algae that can down-regulate their carbon concentrating mechanisms to save 114 energy (Hepburn et al. 2011, Cornwall et al. 2012, Raven et al. 2012).

Coralline algae are thought to be one of the groups of species most vulnerable to ocean 115 116 acidification due to the solubility of their high magnesium-calcite skeletons. Ocean 117 acidification is causing a decrease in the saturation state of calcium carbonate ( $\Omega = [Ca^{2+}] \times$  $[CO_3^{2-}] / K_{sp}$  which is likely to affect the ability of marine calcifiers to form their carbonate 118 119 skeleton or shells by a decline in calcification rates (Kroeker et al. 2010). Although the 120 physiological response in terms of calcification is variable among taxa and species (Ries et al. 121 2009), seawater acidification is related to reduced growth rates in calcified macroalgae 122 (Kroeker et al. 2013). The recruitment and growth of coralline algae are usually negatively affected under elevated  $pCO_2$  (Table 1). Reductions in calcification rate at elevated  $pCO_2$  have 123

- been demonstrated for most coralline algae (Harley et al. 2012) but this response is variable
- 125 among species (Fig. 4, Table 1).
- 126

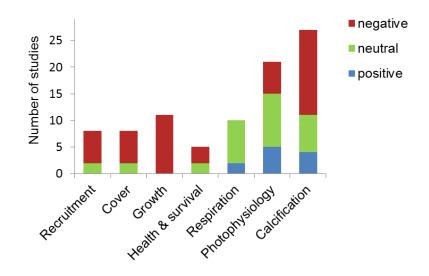


Fig. 4. Summary of the expected direct effects of ocean acidification on coralline algae this century (based on studies in Table 1). Negative, neutral and positive effects correspond to decreases, no change, and

increases in the processes, respectively. Photophysiology includes both photosynthesis and photosystem II
 efficiency and relative electron transport rate.

133

# 134 **4.2.3 Interactive effects of ocean warming and acidification on seaweeds**

135 While the effects of climate change and ocean acidification on seaweeds have mostly been 136 studied in isolation from each other, there is a lack of studies how these factors interact. The 137 combination of these two factors could however have a larger impact than alone. Recent studies 138 have effectively shown that the negative effect of ocean acidification on coralline algal 139 calcification is exacerbated by further ocean warming (Anthony et al. 2008). Some authors reported a significant pCO<sub>2</sub> effect on coralline calcification, health and survival only in 140 combination with increased temperature, suggesting that elevated pCO<sub>2</sub> aggravates the 141 142 sensitivity of coralline algae to temperature (Martin & Gattuso 2009).

143

## 144 **4.2.4** Physiological response of coralline algae to climate change and ocean acidification

145 Photosynthesis, respiration, and calcification in coralline algae are usually determined from measurements of oxygen, DIC and alkalinity in incubation chambers (Fig. 5). Most of these 146 147 studies show that coralline algal calcification is negatively affected under elevated  $pCO_2$  (Table 148 1) and that this effect is exacerbated by warming (Anthony et al. 2008). However, some work 149 only shows a significant  $pCO_2$  effect on calcification when this is combined with an increase in 150 temperature (Martin & Gattuso 2009) and some experiments have shown a positive effect of moderate increases in pCO<sub>2</sub> (Smith & Roth, 1979, Ries et al. 2009, Martin et al. 2013, Kamenos 151 et al. 2013). Such responses may be related to the ability of the algae to maintain an elevated 152 153 pH at the site of calcification despite reduced external pH to facilitate CaCO<sub>3</sub> precipitation 154 (Borowitzka 1987, Ries et al. 2009, Hurd et al. 2011, Cornwall et al. 2013b).



157 Fig. 5. Experimental set-up (a) with replicated aquaria (b) supplied with seawater at ambient or elevated

- 158 pCO<sub>2</sub> containing alive or dead *Lithothamnion glaciale* maerl that were incubated in Perspex chambers for 159 metabolic fluxes analyses (c,e, dead maerl and d,f, live). Photograph by Sophie Martin (Experiments
- 160 performed at Plymouth Marine Laboratory).

161 The response of coralline photosynthesis to increased  $pCO_2$  is also variable among species with 162 negative, neutral, positive and parabolic responses (Table 1) that can depend on the light levels used (Martin et al. 2013). Studies investigating the effect of increased  $pCO_2$  on respiration 163 164 showed no response for most species (Fig. 4). Very few studies of coralline algae have 165 investigated photosynthesis, respiration, and calcification all together yet we know that these processes are complex and tightly linked. Photosynthesis affects calcification through the 166 167 formation of the fibrous organic matrix of the cell walls where the nucleation of calcite crystals 168 is thought to occur (Borowitska, 1981). In addition, both photosynthesis and respiration affect 169 calcification through changes in pH that occur in the cell walls at the site of calcification (Smith 170 & Roth 1979, Gao et al. 1993) but also in the diffusion boundary layer between the algal surface 171 and external seawater (Hurd et al. 2011). Photosynthesis (or respiration) increases (or 172 decreases) pH and thereby increases (or decreases) CaCO<sub>3</sub> saturation state, promoting (or 173 hindering) the precipitation of CaCO<sub>3</sub>. Coralline algae are able to maintain calcification in the 174 dark even at the relatively low pH values generated by respiration. Digby (1977) and Hofmann 175 et al. (2012b) postulate that carbonic anhydrase may also play a role in the calcification of 176 coralline algae by catalysing the conversion of  $CO_2$  into  $HCO_3^-$  and then  $CO_3^{2^-}$ . The stimulation 177 of carbonic anhydrase activity could help prevent a decrease in calcification at elevated  $pCO_2$ 178 as reported for the Mediterranean crustose coralline alga *Lithophyllum cabiochae* (Martin et al. 179 2013). However, carbonic anhydrase is also used by photosynthesis to convert  $HCO_3^-$  to  $CO_2$ . 180 The maintenance or enhancement of calcification rates under elevated  $pCO_2$  in L. cabiochae 181 may thus be detrimental to photosynthesis, as indicated by reduced photosynthesis under

182 elevated  $pCO_2$  (Martin et al. 2013).

183 The physiological response of maerl to warming and ocean acidification has been investigated 184 in Lithothamnion corallioides which is a temperate species (Noisette et al. 2013b) and 185 Lithothamnion glaciale which is a cold-water species (Büdenbender et al. 2011, Kamenos et al. 2013). In L. corallioides diel (24h) calcification decreased by 50% at 750 µatm and 80% at 186 187 1000 µatm, and in L. glaciale the thalli dissolve under elevated pCO<sub>2</sub> (Büdenbender et al. 2011). 188 Kamenos et al. (2013) found that at night L. glaciale calcified in the control treatment but 189 dissolved in the low pH treatment. In both species calcification is less affected by  $pCO_2$  in the 190 light that in the dark, although in L. glaciale upregulated calcification occurs at low pH cf. 191 control treatments during the day (Noisette et al. 2013b, Kamenos et al. 2013). High latitude 192 maerl beds are at risk as surface waters are becoming more corrosive which is expected to cause 193 dissolution of these habitats (Büdenbender et al. 2011). Tropical rhodolith beds may also be 194 affected since in multispecies rhodoliths, made up of *Lithophyllum*, *Hydrolithon* and *Porolithon* 195 spp., there is a decrease calcification by 20-250% between control and acidified conditions 196 (Jokiel et al. 2008, Semesi et al. 2009).

197

## 198 4.2.5 Response of early life stages to ocean acidification and warming

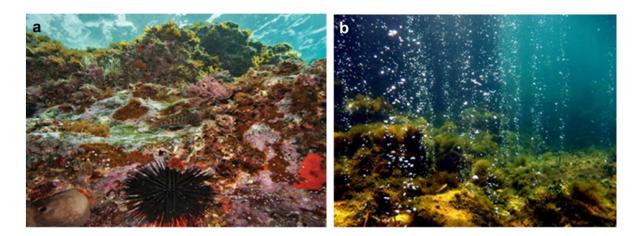
199 Although fragmentation is the main source of new thalli in beds of rhodoliths and maerl, they 200 are initiated through recruitment from spores (Foster 2001). Unfortunately, there is mounting 201 evidence that ocean acidification and warming will have negative impacts upon the 202 recruitment and growth of early life history stages of coralline algae (Agegian 1985, Jokiel et 203 al. 2008, Kuffner et al. 2008, Russell et al. 2009, Porzio et al. 2013, Bradassi et al. 2013; 204 Table 1). Kuffner et al. (2008) found that impacts on settlement led to 90% lower tropical 205 crustose coralline cover at pH 7.9 than at pH 8.2. Some coralline algal species show a 206 reduction in reproductive structures in areas with naturally high  $pCO_2$  conditions (Porzio et al. 207 2011) and germination of spores in the laboratory reveal developmental abnormalities and 208 increased mortality in acidified conditions. Bradassi et al. (2013) found that in acidified 209 conditions the germlings of an intertidal species of coralline algae were able to fight

- 210 dissolution by up-regulating their rates of calcification; this must increase energy costs and
- 211 helps explain why coralline algae are replaced by fleshy algae in naturally high  $pCO_2$
- 212 conditions (Hall-Spencer et al. 2008, Porzio et al. 2013).
- 213

## 214 **4.2.6 Ecosystem level responses**

215 Most research into the effects of ocean warming and acidification has involved relatively 216 short-term (1 year or less) experiments on single species which makes it difficult to scale-up 217 and predict long-term effects at the community and ecosystem levels (Russell et al. 2013). To 218 tackle this, areas with naturally high CO<sub>2</sub> (and/or low pH and low calcium carbonate 219 saturation states) are being used to investigate which organisms can tolerate the long-term 220 consequences of ocean acidification and reveal how communities of primary producers 221 respond (Hall-Spencer et al. 2008, Johnson et al. 2012, Inoue et al. 2013). As with laboratory 222 and mesocosm experiments, the vent systems cannot accurately mimic future ocean 223 conditions; acidified areas are open systems so corallines can recruit from unaffected habitats 224 and grazing fish can swim in and out, these systems typically have large variations in 225 carbonate chemistry, and in some cases there are confounding factors that may mask or 226 amplify the effects of CO<sub>2</sub>. Some efforts have been made to determine the combined effects 227 of warming and acidification at such sites, although such work has so far been restricted to

- corals, molluscs and bryozoans (Rodolfo-Metalpa et al. 2011). Some volcanic vents are
- proving to be particularly useful 'natural laboratories' for the study of ocean acidification as
- they reveal tipping points in recruitment, growth, survival and species interactions along
- $pCO_2$  gradients (Porzio et al. 2011, 2013). Many species of microalgae, macroalgae and
- 232 seagrasses are remarkably tolerant of long-term exposures to high and variable carbon dioxide
- 233 levels at tropical and temperate  $CO_2$  seeps (Johnson et al. 2012, 2013, Russell et al. 2013).
- That they tolerate these conditions does not mean that they will necessarily thrive; seagrasses
- for example lose the ability to defend themselves against herbivores and become over-run by
- competing species of invasive seaweeds (Arnold et al. 2012).
- 237 One clear pattern in studies at multiple natural CO<sub>2</sub> gradients is that the Corallinales are
- adversely affected (Porzio et al. 2011, Fabricius et al. 2011) which is a particular concern
- since this group of organisms underpin maerl, rhodolith habitat production (Hall-Spencer et al. 2010). At Mediterranean CO<sub>2</sub> seeps the Corallinales dissolve at low carbonate saturation
- al. 2010). At Mediterranean  $CO_2$  seeps the Corallinales dissolve at low carbonate saturation states (Martin et al. 2008) and are outcompeted by heterokont algae as  $CO_2$  levels increase
- 242 (Fig. 6). Epilithic diatoms, Dictyotales (e.g. *Dictyota* spp., *Padina* spp.) and Fucales (e.g.
- 243 *Cystoseira* spp., *Sargassum* sp.) thrive as CO<sub>2</sub> levels increase although *Padina* spp. loose
- calcification (Porzio et al. 2011, 2013, Johnson et al. 2012). At the ecosystem level the
- coralligenous habitats loose biodiversity as carbonate levels fall along transects of increasing
- 246 levels of CO<sub>2</sub>; benthic recruitment is disrupted in calcified organisms (Cigliano et al. 2010);
- some organisms such as soft corals and anemones are tolerant but many are physiologically
- unable to cope with the increased  $CO_2$  levels (Suggett et al. 2012, Calosi et al. 2013, Inoue et
- al. 2013). Reefs formed by an association between coralline algae and vermetid molluscs are
- expected to become extinct this century unless  $CO_2$  emissions are reduces and emergency
- conservation measures are taken (Milazzo et al. 2014).
- 252



- 253
- Fig. 6. Sea urchin and coralline algae dominated rocky shore at (a) ambient CO<sub>2</sub> (photograph by David Liittschwager, National Geographic) and (b) the loss of coralline algae and proliferation of diatoms and phaeophytes at a CO<sub>2</sub> seep off Ischia, Italy (photograph by Luca Tiberti, Associazione Nemo).

Additional insights into the community-level effects of warming and ocean acidification are beginning to come from longer-term multispecies laboratory experiments (Hale et al. 2011). These also demonstrate that future increases in CO<sub>2</sub> are likely to have strong negative effects on coralline algae and positive or neutral effects on noncalcified seaweeds both directly via improved growth and indirectly via reduced consumption by calcified herbivores such as sea

urchins (Connell & Russell 2010, Diaz-Pulido et al. 2011, Cornwall et al. 2012, Olischläger et
al. 2012, Roleda et al. 2012, Olabarria et al. 2013). It is clear that the impact of climate

- 264 change and ocean acidification on rhodolith/maerl beds will depend on the combined
- 265 influence of direct environmental impacts on individual species and indirect effects mediated
- by changes in interspecific interactions (Harley et al. 2012). For example, ocean acidification
- 267 may disrupt invertebrate recruitment to maerl/rhodolith beds by altering chemical settlement
- 268 cues associated with crustose coralline algae, as shown with coral recruits (Doropoulos & Diaz Pulido 2013)
- 269 Diaz-Pulido 2013).
- 270 Ocean warming and acidification degrade the habitat complexity of coralline algal sediments
- 271 with profound effects since these algae induce settlement and recruitment of numerous
- invertebrates. The rapid rate of physicochemical change is predicted to cause a shift from
- calcareous to fleshy algal-dominated coastal ecosystems (Hall Spencer et al. 2008, Hoegh-
- Guldberg et al. 2007, Anthony et al. 2011, Diaz-Pulido et al. 2011). Given that
- rhodolith/maerl habitats usually only form in waters that have high carbonate saturation states,
- the spread of low saturation state waters in upwelling areas, lowered salinity water masses and in polar waters, are likely to reduce their habitat complexity and thus cause biodiversity loss.
- 278 Beds of unattached coralline algae provide nursery grounds and habitat for numerous
- commercial species of invertebrates and fishes (Kamenos et al. 2004a,b) so a decline in these
- habitats due to dissolution is also likely to have serious consequences for society and
- economy through the impact on fisheries.
- 282

# 283 4.2.7 Impact on global C and CaCO<sub>3</sub> budget

- 284 Rhodolith/maerl beds are a significant component of carbon and carbonate cycles in shallow
- coastal ecosystems, being major contributors to CO<sub>2</sub> fluxes through high community
- photosynthesis and respiration (Martin et al. 2005, 2007) and through high CaCO<sub>3</sub> production
   and dissolution (Martin et al. 2006, 2007). The habitats formed by coralline algae are
- expected to be degraded by ocean acidification and warming as they have slow growth rates
- and are easily corroded due to their soluble high Mg-calcite skeletons. Any major decline in
- 290 coralline algae would have dramatic consequences since they cover vast areas of rock and can
- form very extensive beds on sediments (Amado-Filho et al. 2012). Changes in the balance
- between algal carbonate production and dissolution induced by elevated  $pCO_2$  and
- 293 temperature in maerl and rhodolith beds may have major implications for carbon dynamics in
- coastal systems by affecting the carbonate chemistry of the water column and the ability of
- 295 the oceans to take up atmospheric  $CO_2$  (Andersson et al. 2005).
- 296 In temperate maerl beds, L. corallioides calcification is expected to decrease by up to 80% at 297 1000  $\mu$  atm pCO<sub>2</sub>, relative to present day conditions of pCO<sub>2</sub> (Noisette et al. 2013b). This 298 would cause a reduction in CaCO<sub>3</sub> precipitation from a current value of ca. 500 g CaCO<sub>3</sub> m<sup>-2</sup>  $y^{-1}$  (Martin et al. 2007) to less than 100 g CaCO<sub>3</sub> m<sup>-2</sup> y<sup>-1</sup> in the near future under 1000 µatm 299 300 pCO<sub>2</sub> (Noisette et al. 2013b). The net calcification in maerl/rhodolith beds could even become 301 negative because the production of CaCO<sub>3</sub> will be exceeded by its dissolution (Martin & 302 Gattuso 2009). Arctic maerl beds are at particular at risk due to seawater becoming corrosive 303 to algal carbonate since live specimens of polar L. glaciale are corroded under elevated  $pCO_2$ 304 (Büdenbender et al. 2011). Dissolution also occurs in the dark in live maerl under elevated 305  $pCO_2$  in temperate maerl species (Noisette et al. 2013b; Kamenos et al. 2013). As maerl beds 306 are also composed of dead thalli, dissolution is likely to strongly increase; dissolution of dead 307 L. glaciale thalli increases by more than 10 fold from ambient condition to elevated  $pCO_2$  of 308 ca. 1100 µatm (Kamenos et al. 2013). Precipitation and dissolution of CaCO<sub>3</sub> contribute to
- 309 the global CO<sub>2</sub> balance through shifts in the seawater carbonate equilibrium. The precipitation

- of one mole of CaCO<sub>3</sub> (Ca<sup>2+</sup> + 2HCO<sub>3</sub><sup>-</sup>  $\leftrightarrow$  CaCO<sub>3</sub> + H<sub>2</sub>O + CO<sub>2</sub>) releases *ca*. 0.6 moles of
- 311  $CO_2$  in seawater (Ware et al. 1992). In that way, the  $CO_2$  released by calcification in
- 312 maerl/rhodolith beds will also be reduced. These changes will thus have major implications
- 313 for both carbon and carbonate budgets in coastal systems.
- 314

## 315 4.2.8 Acclimation and adaptation

316 The coralline fossil record for the past 300 Million years shows they have been able to survive 317 past mass extinctions including periods of very high CO<sub>2</sub> (Wood 1999). However, the present 318 rate of ocean change may be too rapid for genetic adaptation of habitat-forming coralline algae. Work at CO<sub>2</sub> vents that are 100s of years old indicates that corallines have a limited 319 320 capacity to acclimate to ocean acidification, since most species are intolerant of chronic 321 exposures to increases in  $pCO_2$  levels predicted this century and those that can survive are 322 outcompeted by fleshy algae (Martin et al. 2008, Porzio et al. 2011, Fabricius et al. 2012). 323 Localised seeps with high CO<sub>2</sub> levels are not well suited to the study of coralline algal 324 adaptation since they are open to colonization by algal spores from outside the acidified areas 325 and so presently we have no knowledge about the ability of rhodolith/maerl species to adapt 326 to present day rates of warming and acidification. We find it striking, however, that 327 rhodolith/maerl beds are common along the Atlantic seaboard of the Americas, where 328 carbonate saturation states are high, but they are scarce along the Pacific seaboard where

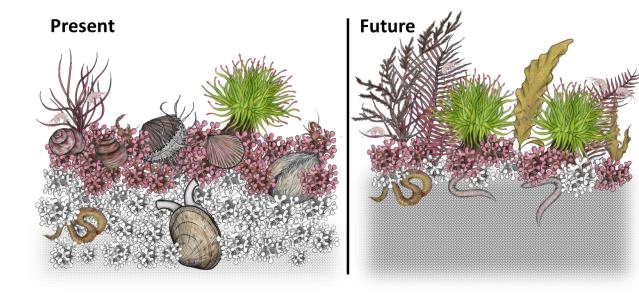
- 329 carbonate saturation states are low.
- Adaptation is the evolutionary response of a population over multiple generations to
- and environmental changes but, according the current magnitude and rate of ocean warming and
- acidification, the potential for evolutionary adaptation is limited in organisms with long
- 333 generation times such as coralline algae. However, coralline algae may have the potential to
- adjust to modified environment within their lifetime (acclimation). In particular, the ability of
- organisms to tolerate significant temperature and/or  $pCO_2$  fluctuations may be a result of
- adaptation (a genetic trait shared by the population) and/or acclimation (owing to phenotypic
   plasticity of the individual). Organisms surviving in highly variable environments are likely to
- be more robust. For instance, Egilsdottir et al. (2013) reported that coralline algae inhabiting
- variable environments where  $pH/pCO_2$  fluctuates naturally are likely to exhibit fewer negative
- responses to elevated  $pCO_2$  than those inhabiting relatively stable environments, supporting
- 341 the assumption of a greater resilience of organisms acclimated and/or adapted to highly
- 342 variable  $pH/pCO_2$  environments of future ocean acidification.
- 343 Coralline algae may be able to tolerate ocean acidification through changes in the composition
- of their skeletons (Agegian 1985, Ries et al. 2009, Egilsdottir et al. 2013). A decrease in
- 345 mMg/Ca ratio would confer resilience to elevated  $pCO_2$  as this would lower the solubility of
- their skeletons. However, it is questionable whether mineralogical plasticity associated with a
- 347 decrease in Mg incorporation will help since warming simultaneously increases Mg
- incorporation (Agegian 1985). Dolomite (MgCO<sub>3</sub>) rich crustose coralline algae have 6- to 10-
- fold lower rates of dissolution than predominantly Mg-calcite species (Nash et al. 2013).
- 350 Dolomite-rich crustose coralline algae are widespread in shallow wave-exposed habitats in the 351 tropics but have not been recorded in cooler waters suggesting that its formation may be
- 352 constrained by temperature. Thus dolomite intracellular calcification may confer an advantage
- 353 on tropical corallines but not those found at higher latitudes (Nash et al. 2011). It is clear that
- elevated  $pCO_2$  weakens the skeletal structure in the high latitude maerl species *L. glaciale*
- 355 (Ragazzola et al. (2012).

- 356 Society can also help mitigate adverse effects of warming and acidification. In the NE
- 357 Atlantic steps have been taken to remove damaging dredging, fishing and aquaculture
- 358 practices to protect maerl beds in a network of protected areas (Chapter ##). If this approach
- is adopted world-wide the reduced pressure on coralline algal systems would help increase
- 360 their resilience to the adverse effects of acidification and warming.
- 361

# 362 **4.3 Conclusions**

363 Beds of unattached coralline algae are scarce in waters with naturally low saturation states of 364 carbonate and are likely to decline in the near future as corrosive waters spread due to ocean acidification. This will lead to a degradation of these ecosystems, reducing habitat 365 366 complexity and their associated biodiversity. Although responses to ocean warming and acidification are variable among species, beds of maerl and rhodoliths are at risk at high 367 latitudes where seawater is becoming corrosive to their high Mg-calcite skeletons. Loss of 368 369 both live and dead maerl is expected to spread to lower latitudes over the century. Work on 370 the effects of ocean warming and acidification indicates that certain fleshy algae and non-

- 371 calcified animals may benefit and outcompete corallines and those calcified animals that
- 372 characterise these habitats (Fig. 7).
- 373 Given the importance of coralline algae, and the scale of the repercussions if they are 374 adversely impacted by rising CO<sub>2</sub> levels, concerted efforts are required to test whether our 375 concerns are justified. More detailed examinations of the physiology/ecology of corallines 376 need to be undertaken as we have an incomplete understanding of how OA and temperature 377 influence gross calcification, growth, dissolution and competitive interactions between 378 seaweeds. Given that society relies upon coralline algae for ecosystems services, such as 379 providing settlement cues or providing habitat for commercially important species, we 380 recommend that biological monitoring programmes begin as soon as possible since changes 381 may be occurring at a rate that will exceed the environmental niches of numerous coralline
- alga taxa, testing their capacities for acclimation and genetic adaptation.
- 383



- Fig. 7. Predicted changes to a typical maerl bed with rising temperature and CO<sub>2</sub> levels. Present day
- 386 maerl beds with alive (above) and dead (below) unattached coralline are inhabited by a rich diversity of 387 calcifying (e.g. gastropods, bivalves, decapods, echinoids) and non-calcifying species. As waters become
- corrosive to carbonate future maerl beds are expected to be degraded, with the loss of habitat complexity
- and biodiversity, although certain fleshy macroalgae and non-calcified fauna (e.g. anemones, polychaetes,
- amphipods) are expected to proliferate (drawing by Sophie Martin).
- 391
- 392 Credits
- 393 IPCC 2013: Summary for Policymakers. In Climate Change 2013: The Physical Science
- 394 Basis. Working Group I Contribution to the Fifth Assessment Report of the Intergovernmental
- Panel on Climate Change, Figure SPM.7 (a) and (c); Figure SPM.8 (a) and (d). [Cambridge
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- 397

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- 404
- 405 **Figure legends**
- 406 **Figure 1.** Equilibrium between the three different species of dissolved inorganic carbon in
- 407 seawater. Atmospheric  $CO_2$  is absorbed at the surface of the oceans where it combines with
- 408 the water molecules to form carbonic acid, which dissociates into bicarbonate, carbonate, and
- 409 hydrogen ions. Different forms of dissolved inorganic carbon (DIC =  $[CO_2] + [HCO_3] + [HCO_3]$
- 410 [CO<sub>3</sub><sup>2-</sup>]) follow thermodynamic equilibria in seawater for the following reactions: CO<sub>2</sub> + H<sub>2</sub>O 411  $\leftrightarrow$  H<sub>2</sub>CO<sub>3</sub>  $\leftrightarrow$  HCO<sub>3</sub><sup>-</sup>  $\leftrightarrow$  H<sup>+</sup> + CO<sub>3</sub><sup>2-</sup> + 2H<sup>+</sup>.
- 412 **Figure 2.** Projected change in global annual mean surface temperature and global mean ocean
- 413 surface pH, relative to 1986-2005 for Representative Concentration Pathway 2.6 (blue) and
- 414 RCP8.5 (red) (Source: IPCC 2013).
- Figure 3. Annual mean temperature and pH changes for the scenarios RCP2.6 and RCP8.5 in
  2081-2100 relative to 1986-2005 (Source: IPCC 2013).
- 417 Figure 4. Summary of the expected impacts of ocean acidification on coralline algae this418 century (based on studies in Table 1).
- 419 **Figure 5.** Experimental set-up (a) with replicated aquaria (b) supplied with seawater at
- 420 ambient or elevated  $pCO_2$  containing alive or dead *Lithothamnion glaciale* maerl that were
- 421 incubated in Perspex chambers for metabolic fluxes analyses (c,e, dead maerl and d,f, live).
- 422 Photo S. Martin (Experiments performed at Plymouth Marine Laboratory).
- 423 Figure 6. Sea urchin and coralline algae dominated rocky shore at (a) ambient CO<sub>2</sub>
- 424 (photograph by David Liittschwager, National Geographic) and (b) the loss of coralline algae

- 425 and proliferation of diatoms and phaeophytes at a CO<sub>2</sub> seep off Ischia, Italy (photograph by
- 426 Luca Tiberti, Associazione Nemo).
- 427 **Figure 7.** Predicted changes to a typical maerl bed with rising temperature and CO2 levels.
- 428 Present day maerl beds with alive (above) and dead (below) unattached coralline are inhabited
- 429 by a rich diversity of calcifying (e.g. gastropods, bivalves, heart urchins) and non-calcifying
- 430 species. As waters become corrosive to carbonate future maerl beds are expected to be
- 431 degraded, with the loss of habitat complexity and biodiversity, although certain fleshy
- 432 macroalgae and non-calcified fauna (e.g. anemones, polychaetes) are expected to proliferate
- 433 (drawing by Sophie Martin).
- 434

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