

2023-05-15

# Ocean acidification stunts molluscan growth at CO<sub>2</sub> seeps

Zhao, L

<http://hdl.handle.net/10026.1/20492>

---

10.1016/j.scitotenv.2023.162293

Science of The Total Environment

Elsevier BV

---

*All content in PEARL is protected by copyright law. Author manuscripts are made available in accordance with publisher policies. Please cite only the published version using the details provided on the item record or document. In the absence of an open licence (e.g. Creative Commons), permissions for further reuse of content should be sought from the publisher or author.*

1 This is the author's accepted manuscript. The final published version of this work (the  
2 version of record) is published by Elsevier in Science of the Total Environment. The  
3 manuscript was made available online on the 13 February 2023 at  
4 <https://www.sciencedirect.com/science/article/pii/S0048969723009099?dgcid=coauthor>  
5 This work is made available online in accordance with the publisher's policies. Please refer to  
6 any applicable terms of use of the publisher.

## 7 **Ocean acidification stunts molluscan growth at CO<sub>2</sub> seeps**

8  
9 **Liqiang Zhao**<sup>a, b, \*</sup>, **Ben P. Harvey**<sup>c, \*</sup>, **Tomihiko Higuchi**<sup>b</sup>, **Sylvain Agostini**<sup>c</sup>, **Kentaro**  
10 **Tanaka**<sup>b</sup>, **Naoko Murakami-Sugihara**<sup>b</sup>, **Holly Morgan**<sup>d</sup>, **Phoebe Baker**<sup>d</sup>, **Jason M.**  
11 **Hall-Spencer**<sup>c, d</sup>, **Kotaro Shirai**<sup>b</sup>

12 <sup>a</sup> *College of Fisheries, Guangdong Ocean University, Zhanjiang 524088, China*

13 <sup>b</sup> *Atmosphere and Ocean Research Institute, The University of Tokyo, Chiba 277-8564,*  
14 *Japan*

15 <sup>c</sup> *Shimoda Marine Research Center, University of Tsukuba, Shimoda 415-0025, Japan*

16 <sup>d</sup> *School of Biological and Marine Sciences, University of Plymouth, Plymouth PL4 8AA,*  
17 *UK*

18 *\* Co-corresponding authors:*

19 *lzhaog@gdou.edu.cn (L.Z.); ben.harvey@shimoda.tsukuba.ac.jp (B.P.H.)*

20 **Abstract**

21 Ocean acidification can severely affect bivalve molluscs, especially their shell  
22 calcification. Assessing the fate of this vulnerable group in a rapidly acidifying ocean  
23 is therefore a pressing challenge. Volcanic CO<sub>2</sub> seeps are natural analogues of future  
24 ocean conditions that offer unique insights into the scope of marine bivalves to cope  
25 with acidification. Here, we used a 2-month reciprocal transplantation of the coastal  
26 mussel *Septifer bilocularis* collected from reference and elevated pCO<sub>2</sub> habitats to  
27 explore how they calcify and grow at CO<sub>2</sub> seeps on the Pacific coast of Japan. We  
28 found significant decreases in condition index (an indication of tissue energy reserves)  
29 and shell growth of mussels living under elevated pCO<sub>2</sub> conditions. These negative  
30 responses in their physiological performance under acidified conditions were closely  
31 associated with changes in their food sources (shown by changes to the soft tissue  
32 δ<sup>13</sup>C and δ<sup>15</sup>N ratios) and changes in their calcifying fluid carbonate chemistry (based  
33 on shell carbonate isotopic and elemental signatures). The reduced shell growth rate  
34 during the transplantation experiment was further supported by shell δ<sup>13</sup>C records  
35 along their incremental growth layers, as well as their smaller shell size despite being  
36 of comparable ontogenetic ages (5-7 years old, based on shell δ<sup>18</sup>O records). Taken  
37 together, these findings demonstrate how ocean acidification at CO<sub>2</sub> seeps affects  
38 mussel growth and reveal that lowered shell growth helps them survive stressful  
39 conditions.

40

41 **Keywords:** Climate change; Bivalves; Calcification; Acclimation; Sclerochronology

## 42 **1. Introduction**

43 Since the Industrial Revolution, the atmospheric concentration of carbon dioxide  
44 ( $\text{CO}_2$ ) has increased at an unprecedented rate from about 280 ppm in 1750 to 410  
45 ppm in 2018, due mainly to human activities such as burning fossil fuels (Blunden et  
46 al., 2018). About one-third of all anthropogenic  $\text{CO}_2$  has entered the ocean, causing  
47 decreases in seawater pH, carbonate ion ( $\text{CO}_3^{2-}$ ) concentration, and saturation state  
48 of calcium carbonate ( $\text{CaCO}_3$ ), and increases in the partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$ ) and  
49 bicarbonate ion ( $\text{HCO}_3^-$ ) concentration (Caldeira and Wickett, 2003). These  
50 fundamental changes in seawater chemistry, known as ocean acidification (OA), can  
51 severely affect most marine bivalves, one of the most ecologically and economically  
52 important taxonomic groups in global marine ecosystems (e.g., Ekstrom et al., 2015;  
53 Leung et al., 2020a; Melzner et al., 2020; Martins et al. 2021; Leung et al., 2022). In  
54 particular, the ability of marine bivalves to calcify, which in turn controls individual  
55 growth, development, defense and survival, is highly sensitive to  $\text{CO}_2$ -driven changes  
56 in seawater carbonate system (e.g., Waldbusser et al., 2014; Thomsen et al., 2017;  
57 Zhao et al., 2017a; Rajan et al., 2021). Hence, obtaining an integrated understanding  
58 of how OA affects the calcification physiology can represent an important step  
59 forward in assessing the fate of marine bivalves in a rapidly acidifying ocean.

60

61 The calcification of marine bivalves can be thought of as a step-by-step process,  
62 and OA can likely affect each of these separate steps. This includes OA substantially  
63 affecting the source and pathway of calcifying substrates such as  $\text{Ca}^{2+}$  and  $\text{HCO}_3^-$  ions

64 transported from ambient seawater to the calcifying front (Lu et al., 2018; Sillanpää  
65 et al., 2018; Zhao et al., 2018a), and the removal of protons generated during the  
66 CaCO<sub>3</sub> precipitation (Zhao et al., 2017b; Liu et al., 2020). Likewise, the synthesis of  
67 organic matrix for the CaCO<sub>3</sub> crystallization and periostracum formation can succumb  
68 under acidified conditions (Hüning et al., 2013; Ramajo et al., 2016). Nevertheless,  
69 continuous *in situ* monitoring of changes at the calcifying front remains an extremely  
70 challenging task, due to the thin extrapallial space located between the inner shell  
71 layer and mantle tissue (Marin et al., 2012; He et al., 2023). Alternatively, a suite of  
72 geochemical proxies archived in shell carbonates hold great potential for estimating  
73 responses to ocean acidification (Levin et al., 2015). When exposed to elevated  $p\text{CO}_2$ ,  
74 for example, the relative contribution of seawater carbon and metabolic carbon to  
75 the calcifying pool can be examined by shell carbon isotopes (Lu et al., 2018; Zhao et  
76 al., 2018a). The amount of boron and uranium incorporated into shells are inversely  
77 related to the pH in the calcifying fluid (Frieder et al., 2014; Zhao et al., 2018b), and  
78 incorporation of sodium into shells can be closely related to activities of ionic  
79 exchangers by which marine bivalves regulate the acid-base status at the site of  
80 calcification (Zhao et al., 2017a, 2017b). While these significant advances have been  
81 made in disentangling processes constraining the calcification sensitivity to OA, as  
82 demonstrated above, little effort has been devoted to integrating how long-term OA  
83 will affect calcification processes when considered together, thus constraining our  
84 ability to assess the extent of the calcification plasticity in response to OA.

85

86 Experiments performed across multiple generations can likely provide a means  
87 to understand the acclimation potential and mechanistic basis of marine bivalves in a  
88 rapidly acidifying ocean. A growing body of transgenerational studies, for example,  
89 demonstrates that marine bivalves (at least species hitherto studied) may have the  
90 ability to respond plastically and acclimate rapidly to OA scenarios projected by the  
91 end of this century (e.g., Ross et al., 2016; Thomsen et al., 2017; Byrne et al., 2020;  
92 Zhao et al., 2020a). However, despite the great potential of transgenerational  
93 experiments in examining whether bivalves can keep up with the pace of OA, their  
94 relatively long generation times (from months to years) and complex life cycles  
95 complicate multigenerational experiments. In addition, almost all transgenerational  
96 experiments have been carried out under simulated conditions. In the face of  
97 multiple stressors and complexities of the nature, whether rapid transgenerational  
98 acclimation can be still maintained remains unknown. Hence, approaches which can  
99 document the acclimation plasticity under environmentally realistic scenarios are in  
100 high demand.

101

102 A space-for-time approach opens a promising avenue in examining the extent to  
103 which marine bivalves have the capacity to acclimate to OA. Synchronic comparisons  
104 of individuals naturally growing along large gradients of  $p\text{CO}_2$  and pH to substitute  
105 space for time, so-called the space-for-time substitution (Reusch, 2014), can likely  
106 capture long-term ecological relevance and more accurately assess the fate of key  
107 species, populations, and ecosystems under projected OA scenarios. Naturally  $\text{CO}_2$ -

108 enriched habitats, especially volcanic CO<sub>2</sub> seeps that can generate consistently acidic  
109 conditions throughout the year, are increasingly used as natural analogues of future  
110 ocean acidification conditions (Feely et al., 2008; Hall-Spencer et al., 2008; Rastrick et  
111 al., 2018). Within the framework of space-for-time substitution, the large predatory  
112 gastropod *Charonia lampas* living in CO<sub>2</sub> seeps off Shikine Island (Japan) and the  
113 widespread Mediterranean mollusc *Hexaplex trunculus* in Vulcano Island CO<sub>2</sub> seeps  
114 (Italy) show little ability to compensate for corrosive effects of seawater acidification  
115 on shell mineralization (Harvey et al., 2016, 2018). Nevertheless, contrasting findings  
116 have recently been observed in molluscs inhabiting CO<sub>2</sub> seeps in the south-western  
117 Pacific, where the herbivorous gastropod *Eatoniella mortoni* produces thicker, more  
118 crystalline, and mechanically resilient shells for compensatory growth (Leung et al.,  
119 2019). These discrepancies among CO<sub>2</sub> seep habitats can most likely be attributed to  
120 differences in duration, frequency, and intensity of acidification (Aiuppa et al., 2021).  
121 To the best of our knowledge, nevertheless, only a few studies have used natural CO<sub>2</sub>  
122 seeps to advance our knowledge of the acclimation of marine bivalves to projected  
123 OA scenarios (e.g., Hahn et al. 2012; Agostini et al. 2018; Martins et al. 2021).

124

125 Japanese volcanic CO<sub>2</sub> seeps, especially those off Shikine Island, have recently  
126 been developed as natural analogues of future oceanic conditions to show how  
127 marine organisms and ecosystems respond and acclimate to OA (e.g., Inoue et al.,  
128 2013; Agostini et al., 2018, 2021; Cattano et al., 2020; Harvey et al., 2018, 2020,  
129 2021a, 2021b). The intertidal rocky shore communities along the coast of Shikine

130 Island are composed predominantly of calcifying species (mussels, oysters, barnacles,  
131 coralline algae, etc.), whose abundance and distribution vary in both space and time.  
132 Surveys using quadrats (25 × 25 cm) showed that the population density of intertidal  
133 mussels near the CO<sub>2</sub> seeps was reduced compared to outside the seeps (from 7 ± 13  
134 individuals to 2 ± 3 individuals; Agostini et al., 2018), indicating that they may be  
135 negatively impacted by elevated pCO<sub>2</sub> conditions and hence sparking our interest in  
136 testing whether, and to what extent, bivalves living near Shikine Island CO<sub>2</sub> seeps  
137 have the ability to acclimate under these naturally acidified conditions.

138

139 Here, we aimed at gaining a better understanding of the acclimation potential of  
140 the widely distributed mussel *Septifer bilocularis* along the coast of Shikine Island to  
141 ocean acidification. Firstly, we performed a 2-month reciprocal transplant of mussels  
142 between an elevated pCO<sub>2</sub> area and an adjacent reference pCO<sub>2</sub> area to assess their  
143 physiological performance. Secondly, we performed ultra-high-resolution temporal  
144 analysis (on a subset of individuals) of shell geochemistry along growth layers to gain  
145 temporal information for their lifespan. Geochemical approaches, such as stable  
146 carbon and nitrogen isotopic and trace elemental analyses which hold great promise  
147 to disentangle calcification responses of marine bivalves to ocean acidification in  
148 temporal context (as reviewed by Zhao et al., 2020a), were employed to obtain an  
149 integrated insights into mechanisms behind mussel calcification and associated  
150 physiological processes. Our findings provide a more comprehensive understanding  
151 about how mussels respond and likely acclimate in a rapidly acidifying ocean.



152

## 153 **2. Materials and methods**

### 154 **2.1. Study area**

155 Shikine Island (34°19'9" N, 139°12'18" E) is a volcanic island located off the Izu  
156 peninsula in Japan. The shore of the island is composed of crustal rocks varying in  
157 size with the tidal range less than 2 m, and near CO<sub>2</sub> seeps the water depth is  
158 approximate 4 m during high tides. As schematically illustrated in Fig. 1, our two low  
159 intertidal sites, elevated pCO<sub>2</sub> area and reference pCO<sub>2</sub> area, have well characterized  
160 carbonate chemistry (e.g., Agostini et al., 2015, 2018; Cattano et al., 2020; Harvey et  
161 al., 2018, 2020, 2021a, 2021b; Witkowski et al., 2019). The carbonate chemistry of  
162 the two sites at the time of the current study are published in Harvey et al. (2020).  
163 Briefly, seawater pH<sub>T</sub>, temperature and salinity of the two sites were recorded  
164 through *in situ* measurements of the subtidal environment using a multisensor  
165 (WQC-24, TOA-DKK, Japan). Total alkalinity samples were collected as discrete  
166 samples and measured by titration (916 Ti-Touch, Metrohm) with 0.1 mol l<sup>-1</sup> HCl.  
167 Seawater carbonate system parameters were calculated from temperature, salinity,  
168 pH<sub>T</sub> and total alkalinity using the software CO2SYS (Pierrot et al., 2006). The  
169 reference pCO<sub>2</sub> area had a mean pH<sub>T</sub> of 8.041 ± 0.067 (SD) and the elevated pCO<sub>2</sub>  
170 area had a mean pH<sub>T</sub> of 7.719 ± 0.095 (SD). The elevated pCO<sub>2</sub> area represents an  
171 end of the present century projection for reductions in pH under the Representative  
172 Concentration Pathway 8.5 scenario (IPCC, 2013), and had the same temperature,  
173 salinity, dissolved oxygen, total alkalinity, nutrients and depth as the reference site

174 (Agostini et al., 2015, 2018; Harvey et al., 2019, 2021a), as well as the same carbon  
175 isotopic ratio of dissolved inorganic carbon (Fig. 1) relative to the reference  $p\text{CO}_2$   
176 area used for comparison. The gas emitted from the seeps is 98%  $\text{CO}_2$  (Agostini et al.,  
177 2015). The annual range of sea surface temperature varies from 14 to 28 °C and  
178 salinity remains constant at ca. 34 over time (Agostini et al., 2018; Harvey et al.  
179 2021a).

180

## 181 **2.2. Study organism**

182 Mytilidae represent some of the most conspicuous marine bivalves within the  
183 intertidal and shallow subtidal zones (Morton, 2019). Species of *Septifer* (Récluz,  
184 1848), along with many warm-water Mytilidae, are far less well studied when  
185 compared to genera such as *Mytilus* and *Perna* (Morton et al. 2020). *Septifer*  
186 *bilocularis* (Linnaeus, 1758), our study organism, is located along the coasts of  
187 mainland Japan, as well as in the Indo-West-Pacific (Higo et al. 1999). Individuals of *S.*  
188 *bilocularis* have a thick, ovally elongate shell that is ventrally concave, dorsally  
189 peaked with a rounded posterior margin with terminal beaks (Morton et al. 2020).  
190 The shell valves are covered by a thick periostracum over fine radially divaricating  
191 lirae (Morton et al. 2020). This species was the dominant mussel within our study  
192 site, and yet little else about its ecology has previously been described. Individuals of  
193 *S. bilocularis* in Hong Kong are subtidal and associated with coral heads (Dudgeon  
194 and Morton, 1982), whereas individuals (shell length of 23–28 mm and body weight  
195 of 2.3–2.5 g) in our study site are found attached by byssal threads to the rocky

196 substratum in the low intertidal zone.

197

### 198 **2.3. Experimental setup**

199 Adults of *S. bilocularis* inhabiting the low intertidal zone in the elevated  $p\text{CO}_2$  and  
200 reference  $p\text{CO}_2$  areas were randomly collected, individually labelled with a reference  
201 number using queen bee marking tags (E.H. Thorne Ltd., UK), and reciprocally  
202 transplanted for two months (mid-April to mid-June 2019) using a two-way  
203 orthogonal experimental design (as schematically illustrated in Fig. 1). This design  
204 therefore includes procedural controls where some individuals were transplanted  
205 back into the same site they were collected from. Redeployment of individuals for  
206 the two-month reciprocal transplantation was achieved by placing them inside cages  
207 (green mesh, 20 × 10 × 4 cm, 1 cm mesh size) which were attached to the substratum  
208 with anchor bolts in locations next to where the mussel individuals were originally  
209 collected in the low intertidal zone (two cages per site) at the same tidal height (with  
210 a maximum tidal depth of around 2.0 m during spring tides). Each cage contained  
211 five individuals collected from reference  $p\text{CO}_2$  conditions and five individuals  
212 collected from elevated  $p\text{CO}_2$  conditions (individuals were randomly assigned to the  
213 four cages). Seawater samples for the stable carbon isotope ratio ( $\delta^{13}\text{C}$ ) analysis were  
214 collected using 20-ml glass vials and poisoned with two drops of the supersaturated  
215 mercury chloride solution to prevent any biological interference.

216

217 Following the *in situ* transplant experiment, two individuals from each site

218 (elevated  $p\text{CO}_2$  and reference  $p\text{CO}_2$ ) were used for high-resolution analysis over their  
219 lifespan to identify ontogenetic age, annual growth rate, and geochemical properties  
220 via stable isotopic analysis (described in greater detail below). These individuals were  
221 selected from the reciprocal transplant experiment, at random (one from each cage),  
222 using only those procedural control individuals that had been collected from, and  
223 transplanted back into the same site.

224

#### 225 **2.4. Sample collection and individual analysis**

226 Following our *in situ* transplant experiment, all specimens were carefully cleaned and  
227 then frozen at  $-20\text{ }^\circ\text{C}$ . Four or five individuals from each of the four treatments of the  
228 reciprocal transplant were randomly sampled and dissected using a scalpel for  
229 analysis. The whole soft tissue of each mussel was washed with Milli-Q water (18.2  
230  $\text{m}\Omega$ ), freeze-dried for 48 h and then weighed, and the shell air-dried and weighed.  
231 The condition index (CI) of each sample was calculated using the equation (1) (Lucas  
232 and Beninger, 1985).

233

$$234 \text{ CI} = \text{dtw}/\text{dsw} \times 100 \dots\dots\dots(1)$$

235

236 where CI is the condition index, and dtw and dsw represent the dry soft tissue  
237 weight (g) and dry shell weight (g), respectively.

238

239 The newly formed shell portion of each mussel was analyzed to examine its daily

240 growth rate. Specifically, mussel shells were sectioned to accurately analyze the rate  
241 of shell growth and thickness of periostracum formation during the experiment. The  
242 right valve of each shell was embedded in epoxy resin and along the axis of  
243 maximum growth two three-millimeter-thick sections were cut using a low-speed  
244 diamond saw (Isomet 1000, Buehler, Lake Bluff, IL, United States). Afterward, both  
245 sections were mounted on glass slides, ground with 70 and 13  $\mu\text{m}$  diamond cup  
246 whetstones, respectively, polished with 0.3  $\mu\text{m}$  alumina powder, and then  
247 ultrasonically washed with Milli-Q water (18.2  $\text{m}\Omega$ ) to remove any adhering particles.  
248 Photographs of polished shell sections were taken using a digital microscope  
249 (KEYENCE VHX-2000) and the amount of newly formed shell portions (as determined  
250 by internal growth bands; Zhao et al., 2019a) and also thickness of corresponding  
251 periostracum were analyzed using an image processing software ImageJ. The average  
252 daily rate of mussel shell growth during the experiment was subsequently computed.

253

254 To perform high-resolution analysis across the lifespan of the mussel, the  
255 mussels subsampled from the reciprocal transplant experiment were used (one  
256 mussel from each cage in the elevated  $p\text{CO}_2$  area, and one mussel from each cage in  
257 the reference  $p\text{CO}_2$  area). We only used individuals collected from and transplanted  
258 back into the same site. These mussels were soaked in 0.5 wt% HCl overnight at room  
259 temperature to prevent shell surface contamination (Zhao et al., 2019b). Carbonate  
260 samples from the left valve of each mussel were then milled along the axis of  
261 maximum shell growth at approximately 0.1-0.3 mm resolution for subsequent

262 geochemical analysis.

263

#### 264 **2.5. Isotopic analysis via EA-IRMS and CF-IRMS**

265 Approximately 2 mg of soft tissue from each mussel was homogenized and weighed  
266 into a tin capsule, tightly folded, and then analyzed using an elemental analyzer (EA;  
267 vario MICRO cube, Elementar, Germany) interfaced to an isotope ratio mass  
268 spectrometer (IRMS; IsoPrime100, IsoPrime, UK) at the Atmosphere and Ocean  
269 Research Institute (AORI), University of Tokyo. Stable carbon and nitrogen isotope  
270 ratios ( $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$ ) were reported related to the standards (Vienna Pee Dee  
271 Belemnite (VPDB) and the atmospheric nitrogen, respectively), and expressed in the  
272 conventional delta ( $\delta$ ) notation in per mil (‰). Measured  $\delta^{13}\text{C}$ ,  $\delta^{15}\text{N}$  and C/N ratio  
273 were calibrated respectively against standards (L-alanine (AZ101SS13), Shoko  
274 Scientific;  $\delta^{13}\text{C} = -19.6$  ‰,  $\delta^{15}\text{N} = 13.7$ , and C/N = 3.0). Throughout the analysis, the  
275 NIST SRM 2976 (National Institute of Standards and Technology, United States) was  
276 used to monitor instrumental conditions. The reproducibility was better than 0.28 ‰  
277 for  $\delta^{13}\text{C}$  (1  $\delta$ ) and 0.10 ‰ for  $\delta^{15}\text{N}$  (1  $\delta$ ).

278

279 To measure stable carbon and oxygen isotope ratios ( $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$ ) of the shell,  
280 approximately 100  $\mu\text{g}$  of carbonate powder milled from the newly formed shell  
281 portion was weighed into a glass vial and then reacted with 100 % phosphoric acid at  
282 72 °C in a helium-flushed borosilicate exetainer for one hour. Liberated  $\text{CO}_2$  gas was  
283 analyzed using a Thermo Finnigan Delta V Plus continuous flow-isotope ratio mass

284 spectrometer (CF-IRMS) coupled to a Thermo Finnigan GasBench II at AORI.  
285 Measured  $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$  data were calibrated against NBS-19 ( $\delta^{13}\text{C} = +1.95 \text{ ‰}$ ;  $\delta^{18}\text{O} =$   
286  $-2.20 \text{ ‰}$ ). Shell  $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$  values were calculated against VPDB, reported in  $\delta$ -  
287 notation and then given as ‰. Profiles of  $\delta^{18}\text{O}_{\text{shell}}$  provide a chronological framework,  
288 with each shell portion placed into a temporal context allowing the ontogenetic age  
289 of each specimen to be determined. By doing so, the implications of ontogeny in the  
290 interpretation of results from reciprocal transplantation can be substantially  
291 eliminated. The reproducibility was better than 0.05 ‰ for  $\delta^{13}\text{C}$  (1  $\delta$ ) and 0.23 ‰ for  
292  $\delta^{18}\text{O}$  (1  $\delta$ ). Seawater  $\delta^{13}\text{C}$  analysis was performed using CF-IRMS, calibrated and  
293 calculated in the same manner as shell  $\delta^{13}\text{C}$  values.

294

## 295 **2.6. Elemental analysis via LA-ICP-MS**

296 Concentrations of boron (B), Sodium (Na), Magnesium (Mg), Phosphorus (P),  
297 Potassium (K), Strontium (Sr), Barium (Ba), and Uranium (U) in polished shell sections  
298 were analyzed by laser ablation inductively coupled plasma mass spectrometry (LA-  
299 ICP-MS) in 'spot' mode at AORI. Three to six discrete spots based on the incremental  
300 length in the newly formed shell portion during the experiment were set in each shell.  
301 Analyses were performed with an excimer-based laser ablation system (NWR-193,  
302 New Wave Research, Fremont, CA, United States) connected to ICP-MS (7700 CS,  
303 Agilent, Tokyo, Japan). The laser was operated at a pulse rate of 10 Hz, pulse energy  
304 of 1.8 mJ, and beam diameter of 100  $\mu\text{m}$ . Helium was used as the carrier gas. Prior to  
305 each ablation backgrounds were analyzed for 7 s, and subsequent ablation time was

306 15 s followed by 25 s wash out. The signal intensity of ions ( $^{10}\text{B}$ ,  $^{23}\text{Na}$ ,  $^{24}\text{Mg}$ ,  $^{31}\text{P}$ ,  $^{39}\text{K}$ ,  
307  $^{43}\text{Ca}$ ,  $^{88}\text{Sr}$ ,  $^{138}\text{Ba}$ ,  $^{238}\text{U}$ ) was calculated by subtracting the background level of trace  
308 elements empirically obtained with the 0 % laser output set.  $^{43}\text{Ca}$  was used as  
309 internal standard and the element-to- $^{43}\text{Ca}$  ratio was computed and compared to the  
310 standard reference material signal intensity. NIST SRM 612 (synthetic glass) was used  
311 as an external standard, and giant clam powder Jct-1 and coral powder Jcp-1  
312 (National Institute of Advanced Industrial Science and Technology, Tsukuba, Japan)  
313 were measured as quality control materials to confirm accuracy. Calculations on trace  
314 element-to-calcium ratio were based on analyses of NIST SRM 612. To monitor the  
315 instrumental drift, all reference materials were analyzed after every 10-20 samples.

316

## 317 **2.7. Statistical analysis**

318 The  $\delta^{13}\text{C}$  composition of bivalve shells is controlled by seawater  $\delta^{13}\text{C}_{\text{DIC}}$ , the  $\delta^{13}\text{C}$  of  
319 metabolically generated  $\text{CO}_2$ , as well as the proportion of metabolic  $\text{CO}_2$   
320 incorporated into shell carbonate (McConnaughey and Gillikin, 2008). Hence,  
321 determination of the relative contribution of metabolic carbon can be achieved  
322 according to the following equation (2) described by McConnaughey et al. (1997).

323

$$324 \quad \%C_M = (\delta^{13}\text{C}_{\text{shell}} - \epsilon_{\text{cal-b}} - \delta^{13}\text{C}_{\text{DIC}}) / (\delta^{13}\text{C}_M - \delta^{13}\text{C}_{\text{DIC}}) \times 100 \quad (2)$$

325

326 where  $\%C_M$  is the percentage of metabolic carbon incorporated into shell carbonate.

327  $\delta^{13}\text{C}_{\text{shell}}$ ,  $\delta^{13}\text{C}_{\text{DIC}}$  and  $\delta^{13}\text{C}_M$  represent the  $\delta^{13}\text{C}$  of shell carbonate, seawater dissolved



328 inorganic carbon (DIC) and metabolic carbon, respectively, and  $\epsilon_{\text{cal-b}}$  represents the  
329 enrichment factor between calcite and bicarbonate (1 ‰ calculated by Romanek et  
330 al., 1992). In the present study, the  $\delta^{13}\text{C}$  ratios of mussel soft tissues were employed  
331 to approximate  $\delta^{13}\text{C}_M$  according to McConnaughey et al. (1997).

332

333 All data generated following reciprocal transplantation were analyzed using the  
334 IBM SPSS Statistics version 19.0 (SPSS Inc., Chicago, IL, USA). Shapiro-Wilk's test and  
335 Levene's F-test were performed to examine the normal distribution (normality) and  
336 equal variance among groups (homoscedasticity) of experimental data. A fixed-  
337 effects analysis of variance (ANOVA) model of  $p\text{CO}_2$  (2 levels, reference  $p\text{CO}_2$  and  
338 elevated  $p\text{CO}_2$ ) and origin (2 levels, reference  $p\text{CO}_2$  and elevated  $p\text{CO}_2$ ) was used for  
339 the analysis. Following the two-way ANOVA, a Fisher's Least Significant Difference  
340 (LSD) post-hoc test was performed to test the significance of experimental data  
341 among groups designated according to the two-way orthogonal experimental design.  
342 Statistical significance was set at  $p < 0.05$ . The fixed-effects model was initially  
343 compared to a mixed-effects model using the 'transplant cage' as a random-effect for  
344 each endpoint measurement. In all cases, the fixed effects model did not significantly  
345 differ from the mixed effects model and demonstrated lower AIC scores. As such, the  
346 more parsimonious fixed-effects model was employed.

347

### 348 **3. Results**

#### 349 **3.1. Growth performance following transplantation**

350 No mortality was recorded over the course of the transplant experiment. No  
351 statistically significant differences in physiological traits were observed for mussels  
352 transplanted within the same areas (i.e., elevated  $p\text{CO}_2$  or reference  $p\text{CO}_2$   
353 conditions), irrespective of their origin under elevated or reference  $p\text{CO}_2$  conditions  
354 ( $p > 0.05$ ; Figs. 2a and 2b). Residing in elevated  $p\text{CO}_2$  conditions significantly affected  
355 the physiological performance of the mussels, as best exemplified by significant  
356 decreases in their condition index ( $p < 0.05$ ; Fig. 2a) and shell growth ( $p < 0.05$ ; Fig.  
357 2b). Nevertheless, neither  $p\text{CO}_2$  conditions nor the reciprocal transplantation, and  
358 their interaction, significantly affected the thickness of mussel shell periostracum  
359 ( $p > 0.05$ ; Fig. 2c).

360

### 361 **3.2. Somatic stable isotopic signatures following transplantation**

362 Mussels collected under reference  $p\text{CO}_2$  conditions showed significantly higher  
363  $\delta^{13}\text{C}_{\text{soft tissue}}$  values than those inhabiting elevated  $p\text{CO}_2$  conditions ( $p < 0.05$ ; Fig. 3a).  
364 Mussel  $\delta^{15}\text{N}_{\text{soft tissue}}$  signatures were only significantly affected by reciprocal  
365 transplantation, with values significantly higher in mussels collected from reference  
366  $p\text{CO}_2$  areas than those typically residing in elevated  $p\text{CO}_2$  areas when experimentally  
367 exposed to elevated  $p\text{CO}_2$  ( $p < 0.05$ ; Fig. 3b). Mussel C/N ratios were not significantly  
368 affected by either  $p\text{CO}_2$  conditions or reciprocal transplantation ( $p > 0.05$ ; Fig. 3c).

369

### 370 **3.3. Shell geochemical properties following transplantation**

371 While the shells of mussels collected from reference  $p\text{CO}_2$  areas had consistently

372 lower  $\delta^{13}\text{C}_{\text{shell}}$  values in comparison with those in elevated  $p\text{CO}_2$  areas (Fig. 4a), no  
373 significant effect of reciprocal transplantation was found ( $p > 0.05$ ).  $\delta^{13}\text{C}_{\text{shell}}$  ratios  
374 were also not significantly affected by  $p\text{CO}_2$  conditions ( $p > 0.05$ ). Interaction  
375 between these two factors exhibited significant effects on  $\delta^{18}\text{O}_{\text{shell}}$  ratios ( $p < 0.05$ ;  
376 Fig. 4b). Under acidified conditions, mussels collected from reference  $p\text{CO}_2$  areas  
377 showed significantly lower  $\delta^{18}\text{O}_{\text{shell}}$  values than those previously inhabiting elevated  
378  $p\text{CO}_2$  areas ( $p < 0.05$ ). Furthermore, as shown in Fig. 4c, seawater DIC was  
379 determined as the predominant source of shell carbonate, and its contribution to the  
380 DIC pool at the site of calcification increased with increasing  $p\text{CO}_2$  concentration.

381

382 Element-to-calcium ratios of mussel shells following transplantation are shown  
383 in Fig. 5. With increasing  $p\text{CO}_2$  concentration, the amount of B (Fig. 5a) and U (Fig. 5h)  
384 incorporated into shell carbonate decreased significantly ( $p < 0.05$ ). Nevertheless,  
385 the incorporations of Na (Fig. 5b), Mg (Fig. 5c), P (Fig. 5d), K (Fig. 5e), Sr (Fig. 5f), and  
386 Ba (Fig. 5g) were not significantly affected by either  $p\text{CO}_2$  conditions or the reciprocal  
387 transplantation ( $p < 0.05$ ).

388

### 389 **3.4. Analysis of shell geochemical properties over the lifespan**

390 As illustrated in Fig. 6a, high-resolution  $\delta^{18}\text{O}_{\text{shell}}$  profiles of mussels originally living  
391 under reference  $p\text{CO}_2$  conditions and under elevated  $p\text{CO}_2$  conditions were extracted  
392 to constrain the ontogenetic age and rates of annual shell growth. Two mussels  
393 randomly selected from the reference  $p\text{CO}_2$  areas were 7-years-old and 5-years-old,

394 and those living in the elevated  $p\text{CO}_2$  areas were 7-years-old and 6-years-old,  
395 respectively. Following placing each shell portion into a precise context, it is evident  
396 that across their entire lifespan mussels inhabiting elevated  $p\text{CO}_2$  conditions grew at  
397 slower rates in comparison with those living under reference  $p\text{CO}_2$  conditions (Fig.  
398 6b).  $\delta^{13}\text{C}_{\text{shell}}$  profiles provide consistent evidence of lower  $\delta^{13}\text{C}_{\text{shell}}$  values in mussels  
399 living under reference  $p\text{CO}_2$  conditions than those under elevated  $p\text{CO}_2$  conditions  
400 (Fig. 6c).

401

#### 402 **4. Discussion**

403 This is the first investigation of calcification responses of marine bivalves living  
404 around Japanese volcanic  $\text{CO}_2$  seeps. Our observations of significantly lowered shell  
405 growth with increasing elevated  $p\text{CO}_2$  indicate that mussels inhabiting elevated  $p\text{CO}_2$   
406 conditions have their shell formation negatively impacted and do not appear to be  
407 capable of rapidly acclimating to ocean acidification scenarios projected by the end  
408 of this century. These observations are in line with findings of drastically decreased  
409 abundance of bivalves with increasing levels of  $p\text{CO}_2$  along the shore of Shikine Island  
410 (Agostini et al., 2018; Hall-Spencer et al., 2022). Similar observations have also been  
411 made at  $\text{CO}_2$  seeps off islands of São Miguel and Faial in the North Atlantic Ocean,  
412 where the abundance, size and net-calcification of the bivalve (*Ervilia castanea*) were  
413 inversely related to  $p\text{CO}_2$  levels (Martins et al., 2021). Yet, it is worth noting here that  
414 the declines in shell formation under elevated  $p\text{CO}_2$  conditions at  $\text{CO}_2$  seeps are at  
415 odds with previous conclusions that demonstrated marine bivalves inhabiting

416 naturally CO<sub>2</sub>-enriched habitats, such as upwelling areas, can maintain the ability to  
417 calcify by implementing compensatory mechanisms to mitigate effects of elevated  
418 pCO<sub>2</sub> on shell calcification (e.g., Thomsen et al., 2017; Zhao et al., 2018b). In addition,  
419 a growing body of transgenerational experiments has shown that a suite of  
420 physiological processes underpinning the shell calcification can respond plastically  
421 and acclimate rapidly to near-future OA scenarios (e.g., Fitzer et al., 2014; Parker et  
422 al., 2015; Zhao et al., 2017b, 2018a). Given that volcanic CO<sub>2</sub> seeps can be more  
423 reliably used as natural analogues of future oceanic conditions than many other  
424 naturally CO<sub>2</sub>-enriched systems (Rastrick et al., 2018), discrepancies among studies  
425 deserve further attention. Hence, the elucidation of the underlying mechanisms  
426 would represent a key leap forward in our understanding of how bivalves will  
427 respond and acclimate in a rapidly acidifying ocean.

428

#### 429 **4.1. Changes in dietary regimes of mussels under elevated pCO<sub>2</sub> conditions**

430 Following reciprocal transplantation, the condition index of the mussels, which is  
431 widely used as an indication of the energy budget of bivalves (Lucas and Beninger,  
432 1985), significantly decreased for those individuals held under the elevated pCO<sub>2</sub>  
433 conditions, indicating decreases in their energy reserves. Reduced energy reserves in  
434 mussels under acidified conditions can partly be explained by changes in their dietary  
435 regimes. For the first time, the present study utilized  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  ratios of mussel  
436 soft tissues as dietary tracers and revealed marked shifts in food sources of mussels  
437 living in reference and elevated pCO<sub>2</sub> areas. Significant changes in mussel  $\delta^{13}\text{C}_{\text{soft tissue}}$

438 and  $\delta^{15}\text{N}_{\text{soft tissue}}$  demonstrate changes in their principal food sources, which largely  
439 stem from algal detritus (Zhao et al., 2013). Evidence in support of this interpretation  
440 further comes from recent surveys around the same set of  $\text{CO}_2$  seeps off Shikine  
441 Island by Harvey et al. (2021a), according to which diatom algal turfs (*Biddulphia*  
442 *biddulphiana*) largely replaced seaweeds and became the dominant habitat-forming  
443 species around the  $\text{CO}_2$  seeps, thereby potentially affecting components of organic  
444 detritus. Given significant decreases in condition index and thereby energy reserves,  
445 these shifts in food sources are likely much less beneficial for mussel calcification in  
446 response to long-term elevated  $p\text{CO}_2$  stress.

447

448         Nevertheless, it is worth noting here that only the combination of both quality  
449 and quantity of food could reliably dictate dietary regimes of mussels at  $\text{CO}_2$  seeps.  
450 Without doubt, in the present study food sources of mussels varied as  $p\text{CO}_2$  levels  
451 rose, but it remains largely unknown how the nutritional quality of food (which can  
452 be expressed in terms of C:N ratios) responded. At a  $\text{CO}_2$  shallow vent system off  
453 Ischia in the Mediterranean Sea, lowered C:N ratios of organic matter sources with  
454 increasing  $p\text{CO}_2$  levels can likely indicate higher nutritional quality of food produced  
455 by primary producers (Ricevuto et al., 2015). Also, the dominant calorie-enriched turf  
456 algae near  $\text{CO}_2$  seeps in New Zealand had lower C:N ratio (and thus higher nutritional  
457 quality) than that of seaweeds outside seeps (Leung et al., 2018, 2020). Presumably,  
458 in the present study mussels might even consume higher nutritional quality of food  
459 sources under acidified conditions, but evidently such an increase in the quality of

460 food could still be insufficient for compensating against calcification declines.

461

462 Feeding is fundamental for marine bivalves to acquire energy, and its plasticity is  
463 therefore a prerequisite for acclimation in a rapidly acidifying ocean. As reviewed by  
464 Clements and Darrow (2018), many bivalve species experienced depressed feeding  
465 rates and thus reduced energy acquisition when exposed to elevated  $p\text{CO}_2$ . Yet, little  
466 information is available about the role played by the food availability, mainly because  
467 most laboratory-based experiments are carried out under the same feeding regimes.  
468 In the field, previous studies within upwelling systems have demonstrated that food  
469 availability could be sufficient for mussels to maintain shell growth and tissue energy  
470 reserves, while still fulfilling increased energy demands under acidified conditions  
471 (e.g., Parker et al., 2015; Zhao et al., 2018b, 2019a; Ramajo et al., 2020). At  $\text{CO}_2$  seeps  
472 off Shikine Island, while ocean acidification resulted in marked shifts in ecosystems  
473 dominated by turf algae (Harvey et al., 2021a), due to the  $\text{CO}_2$  fertilization, turf algae  
474 can be more likely to rapidly proliferate than macroalgae outside seeps (Harvey et al.,  
475 2021b). The availability of food stemming from predominant turf algal detritus for  
476 mussels living around  $\text{CO}_2$  seeps might be particularly abundant. The role played by  
477 turf algae in the quality and availability of food for mussel consumption deserves  
478 further investigation.

479

480 Taken together, considering significant decreases in condition index and shell  
481 growth, it is reasonable to assume that evident shifts in dietary regimes which may

482 be beneficial for energy acquisition are not sufficient to fully compensate for the  
483 impact of OA on mussel shell formation. As demonstrated below, it is likely that a re-  
484 partitioning of energy amongst different shell calcification processes may occur.

485

#### 486 **4.2. Mechanisms of mussel calcification at CO<sub>2</sub> seeps are less energetically efficient**

487 Understanding the sensitivity of calcification to OA requires an appreciation of the  
488 regulation of calcifying fluid carbonate chemistry. Identification of sources and  
489 pathways of dissolved inorganic carbon (DIC) at the calcifying front has greatly  
490 improved our knowledge in this regard. Up to 100% of shell carbonate was  
491 composed of seawater DIC suggesting that mussels almost exclusively extract  
492 external seawater DIC rather than internal metabolic CO<sub>2</sub> to calcify. While this finding  
493 agrees well with the theoretical calculation of the respiratory gas exchange model,  
494 whereby seawater DIC contributes more than 90% of the calcifying fluid DIC pool  
495 (McConnaughey et al., 1997), evidence is accumulating that marine bivalves,  
496 especially under OA stress, can preferentially transport metabolic CO<sub>2</sub> as a more  
497 efficient and less costly compensatory mechanism to regulate the carbonate  
498 chemistry of their calcifying fluid (e.g., Gillikin et al., 2007; Zhao et al., 2018a; Lu et  
499 al., 2019; Lee et al., 2021). Up to 61% of metabolic carbon, for example, was  
500 observed in shells of Manila clams (*Ruditapes philippinarum*) which exhibited rapid  
501 acclimation to high *p*CO<sub>2</sub> following transgenerational exposure (Zhao et al., 2018a).  
502 Considering that cellular membranes are highly CO<sub>2</sub> permeable (Gutknecht et al.,  
503 1977), metabolic CO<sub>2</sub> can passively diffuse into the calcifying fluid, where it is quickly



504 transformed to  $\text{HCO}_3^-$  through the hydration of carbonic anhydrase (Zhao et al.,  
505 2020b). In comparison to the passive diffusion of metabolic  $\text{CO}_2$ , the transport of  
506 seawater DIC for calcification is thought to be energetically expensive, since the  
507 active transport of  $\text{HCO}_3^-$  (the most likely calcification-relevant DIC species in  
508 seawater) from the ambient seawater to the calcifying front requires many ion  
509 channels, exchangers, and transporters (as reviewed in Zhao et al., 2020a). The  
510 reason that mussels inhabiting the elevated  $p\text{CO}_2$  conditions have mainly employed  
511 mechanisms for active transport of seawater DIC rather than less costly preferential  
512 uptake of metabolic  $\text{CO}_2$  as calcifying substrates remains unknown, but it is intriguing  
513 that the latter usually is seen in bivalves inhabiting intertidal zones where they  
514 habitually experience aerial exposure during the low tide and therefore are used to  
515 counter respiratory acidosis caused by the retention of metabolic  $\text{CO}_2$  (Booth et al.,  
516 1984; Burnett, 1988). Since the mussels used here were located in the very low  
517 intertidal, it is possible that metabolic  $\text{CO}_2$  production during the relatively short  
518 aerial exposure is insufficient to maintain calcification rates, necessitating the use of  
519 the more energetically-costly seawater-derived  $\text{HCO}_3^-$ .

520

521 Active removal of protons generated during the  $\text{CaCO}_3$  precipitation at the site  
522 of calcification is a fundamental requirement for stress tolerance and acclimation to  
523 OA. With increasing  $p\text{CO}_2$  concentrations, the amounts of pH-sensitive elements such  
524 as B and U incorporated into mussel shells decreased significantly, demonstrating  
525 declines of pH at the site of calcification. Similar results have been documented in a

526 variety of coastal bivalves acutely exposed to OA scenarios projected by the end of  
527 this century (e.g., Heinemann et al., 2012; Frieder et al., 2014; Liu et al., 2015; Zhao  
528 et al., 2018b). Perhaps most notably, virtually unaffected  $\text{Na}/\text{Ca}_{\text{shell}}$  and  $\text{K}/\text{Ca}_{\text{shell}}$  ratios  
529 demonstrate that mussels may be unable to complement a suite of less costly and  
530 more efficient ion-regulatory machineries equipped in the mantle tissue, especially  
531  $\text{Na}^+/\text{H}^+$  exchanger (Zhao et al., 2017a; Ramesh et al., 2018), for the removal of  
532 excessive protons in the calcifying fluid. If this is the case, then one would expect  
533 that energy-requiring  $\text{H}^+$ -ATPase may play a major role in actively pumping protons  
534 out of the site of calcification. Taken together, it is reasonable to assume that mussels  
535 collected within the elevated  $p\text{CO}_2$  areas do not possess compensatory mechanisms  
536 that are energetically efficient in the maintenance of calcifying fluid acid-base  
537 homeostasis under OA stress.

538

539       Considering the corrosive nature of acidified waters within the elevated  $p\text{CO}_2$   
540 areas, it is also imperative to account for the degree to which shell calcification could  
541 be counteracted by dissolution and erosion. It has been documented that gastropods,  
542 *Charonia lampas*, collected at our study sites exhibited serious shell dissolution and  
543 visible deterioration (Harvey et al., 2018), due to loss of periostracum – an external  
544 organic cover protecting shells from corrosive seawater (Saleuddin and Petit, 1983).  
545 When maintained under elevated  $p\text{CO}_2$  conditions for two months, mussels did not  
546 lose the thickness of periostracum in the newly formed shell portion and undergo  
547 progressive shell dissolution and erosion, indicating that their shell properties may

548 not be as susceptible to seawater acidification as *C. lampas*. Hence, the resilience of  
549 marine bivalves to calcify in an acidifying ocean also depends on their ability to  
550 counter periostracum loss and shell dissolution. The  $\delta^{13}\text{C}$  analysis indicates that the  
551 formation of mussel periostracum is mediated by organic matrix synthesized and  
552 secreted from the outer mantle epithelium (Zhao et al., 2019b), the latter known to  
553 be extremely energetically expensive and equivalent to 60% of the energy demands  
554 of somatic growth or 150% of the energetic demands of gamete development  
555 (Palmer, 1992). The maintenance of periostracum formation thus indicate that under  
556 acidified conditions mussels give priority to the shell defence rather than calcification,  
557 especially considering that accumulating evidence suggests that the total amount of  
558 energy allocated for shell formation declines considerably under acidified conditions,  
559 (Waldbusser et al., 2013; Li et al., 2016; Spalding et al., 2017; Leung et al., 2020b).

560 Compensatory mechanisms such as the production of alternative calcium carbonate  
561 polymorphs (e.g. Leung et al. 2017) or the production of thicker shells (e.g. Cross et  
562 al. 2019) may additionally help to counteract the negative effects of ocean  
563 acidification on the shell morphology. By combining findings obtained from reciprocal  
564 transplantation (i.e., the limited energy budget, the more energetically costly  
565 calcification processes, and the importance of the periostracum in impeding shell  
566 dissolution), it is, therefore, reasonable to assume that mussels may be forced to  
567 employ functional trade-offs in their energy expenditure in order to survive under  
568 elevated  $p\text{CO}_2$  conditions.

569

570 **4.3. Mussels shrink to survive under elevated  $p\text{CO}_2$  conditions**

571 While most calcifiers (oysters, decapods, gastropods, barnacles, etc.) had significant  
572 decreases in abundance around the volcanic  $\text{CO}_2$  seeps off Shikine Island (Agostini et  
573 al., 2018), patchily distributed dense clusters of mussels were seen along the  
574 gradients of  $\text{CO}_2$ , sparking interest in understanding why mussels can survive OA.  
575 Limited energy budget, less energy-efficient calcification, and sustained periostracum  
576 integrity are traits being exhibited by the mussels following their exposure to  
577 elevated  $p\text{CO}_2$  conditions (Thomsen et al., 2017; Zhao et al., 2017a), which however  
578 cannot satisfactorily explain why mussels can survive at  $\text{CO}_2$  seeps. High-resolution  
579 analysis of shell geochemistry also indicates that mussel calcification is adversely  
580 affected by high  $p\text{CO}_2$ . Evidence in support of this assumption comes from the  
581 consistently higher values of  $\delta^{13}\text{C}_{\text{shell}}$  profiles than those observed in control mussels,  
582 indicating that mussels living near  $\text{CO}_2$  seeps incorporate higher amount of seawater  
583 inorganic carbon (which had higher  $\delta^{13}\text{C}$  values than that of metabolic carbon) to the  
584 site of calcification than those grown outside seeps. Hence, mussels collected from  
585 the elevated  $p\text{CO}_2$  areas adopted energetically costly strategies (e.g., active transport  
586 of seawater DIC rather than the passive diffusion of metabolic carbon) to calcify. The  
587 integrity of periostracum also suggests that energy is being allocated for its  
588 maintenance, as demonstrated above. In line with findings following reciprocal  
589 transplantation, therefore, it is reasonable to assume that mussels suffer a reduced  
590 energy budget in acidified conditions.

591

592 Surprisingly, our chronological findings indicate that, despite their small size  
593 (shell length 1.86 to 2.83 cm), mussels randomly chosen for reciprocal  
594 transplantation were up to 7-years-old. Mussels collected near the CO<sub>2</sub> seeps were  
595 not only older, but also grew slower than those living outside the CO<sub>2</sub> seeps, leading  
596 to the shell length of mussels being smaller by as much as 20%. Presumably, slow  
597 shell building is a strategy used by mussels to survive elevated pCO<sub>2</sub> conditions. A  
598 similar observation has been made for two gastropod species at natural CO<sub>2</sub> seeps  
599 off Vulcano Island, Italy, whereby the ability to adapt through dwarfing conferred a  
600 physiological advantage for the smaller individuals to survive under increased CO<sub>2</sub>  
601 levels (Garilli et al. 2015).

602

603 Slow shell building can substantially lessen the energy burden of stressed  
604 mussels. Similar conclusions have been often drawn on the early life stages of  
605 brooding marine bivalves (e.g., Noisette et al., 2014; Lucey et al., 2015; Waldbusser  
606 et al., 2016). The rates of calcification and energy consumption in the larvae of  
607 brooding oysters *Ostrea lurida* for example are nearly 10 and 50 times slower than  
608 those of broadcast spawning oysters *Crassostrea gigas* (Waldbusser et al., 2016).  
609 Mechanisms underpinning slow shell building as a possible trait for resistance to OA  
610 are suggestively associated with the kinetic-energetic constraint on shell calcification  
611 (Waldbusser et al., 2013, 2014). Reductions in body length to decrease energy needs  
612 when under harsh conditions has been observed in many organisms (Wikelski and  
613 Thom, 2000; Sheridan and Bickford, 2011; Garilli et al. 2015). Smaller body size can

614 be an advantage as  $p\text{CO}_2$  levels continue to rise.

615

616 The use of  $\text{CO}_2$  seeps in this study allowed us to investigate the consequences of  
617 future ocean acidification on mussel shell formation. It should be acknowledged,  
618 however, that this is in the absence of concurrent ocean warming (Hughes et al.  
619 2017), and changes in temperature will mediate the response of organisms and  
620 communities to future ocean acidification. It may be possible in future studies, for  
621 example, to carry out studies that manipulate temperature along  $\text{CO}_2$  gradients (e.g.,  
622 Alessi et al., 2019). Regardless,  $\text{CO}_2$  seeps and other natural analogues are still  
623 invaluable for assessing the future state of organisms, communities, and ecosystems  
624 to future ocean acidification (Rastrick et al., 2018).

625

## 626 **5. Conclusions**

627 This study is the first to provide a comprehensive understanding of how mussels  
628 calcify and grow at  $\text{CO}_2$  seeps off Shikine Island. We observed that mussels inhabiting  
629 elevated  $p\text{CO}_2$  conditions exhibited marked shifts in their food sources, and displayed  
630 significant decreases in their condition index and hence tissue energy reserves. While  
631 the mussels were able to survive, their shell formation was negatively impacted by  
632 ocean acidification, leading to reduced growth and size. They survived through slow  
633 shell building, demonstrated by high-resolution analysis of shell geochemistry over  
634 their lifespan. This slow shell growth is a plastic response that allows them to  
635 maintain a smaller body size, a physiological advantage to enhance energy efficiency.

636 These findings reveal future trajectories for marine bivalves in an acidifying ocean  
637 and provide insights into their scope for acclimation.

638

### 639 **Acknowledgments**

640 We thank the technical staff at Shimoda Marine Research Center, University of  
641 Tsukuba for their assistance aboard RV Tsukuba II and at the study site, and the Japan  
642 Fisheries agencies of Nijima/Shikine Island (Tokyo prefecture) for their support. This  
643 project was supported by the 'International Education and Research Laboratory  
644 Program', University of Tsukuba, and contributes towards the International CO<sub>2</sub>  
645 Natural Analogues (ICONA) Network. This work was also supported by JSPS KAKENHI  
646 Grant Number 17K17622 and 17P17333, and we also acknowledge funding support  
647 from the Ministry of Environment, Government of Japan (Suishinhi: 4RF-1701), the  
648 National Natural Science Foundation of China (42076121, M-0163, 42211530423),  
649 the Guangdong Zhujiang Talents Program (2021QN02H665), the Department of  
650 Education of Guangdong Province (2020KTSCX050, 2022ZDZX4012) and the program  
651 for Scientific Research Start-up Funds of Guangdong Ocean University.

652

### 653 **Data Availability Statement**

654 The data that support the findings of this study are openly available in PANGAEA.

655

### 656 **References**

657 Agostini, S., Harvey, B.P., Wada, S., Kon, K., Milazzo, M., Inaba, K., Hall-Spencer, J.M.,

658 2018. Ocean acidification drives community shifts towards simplified non-  
659 calcified habitats in a subtropical-temperate transition zone. *Sci. Rep.* 8, 111354.

660 Agostini, S., Houlbrèque, F., Biscéré, T., Harvey, B. P., Heitzman, J. M., Takimoto, R.,  
661 Yamazaki, W., Milazzo, M., Rodolfo-Metalpa, R., 2021. Greater mitochondrial  
662 energy production provides resistance to ocean acidification in “winning”  
663 hermatypic corals. *Front. Mar. Sci.* 7, 1235.

664 Agostini, S., Wada, S., Kon, K., Omori, A., Kohtsuka, H., Fujimura, H., Tsuchiya, Y., Sato,  
665 T., Shinagawa, H., Yamada, Y., Inaba, K., 2015. Geochemistry of two shallow CO<sub>2</sub>  
666 seeps in Shikine Island (Japan) and their potential for ocean acidification research.  
667 *Reg. Stud. Mar. Sci.* 2, 45-53.

668 Aiuppa, A., Hall-Spencer, J. M., Milazzo, M., Turco, G., Caliro, S., Di Napoli, R., 2021.  
669 Volcanic CO<sub>2</sub> seep geochemistry and use in understanding ocean acidification.  
670 *Biogeochemistry* 152, 93-115

671 Alessi, C., Giomi, F., Furnari, F., Sarà, G., Chemello, R., Milazzo, M., 2019. Ocean  
672 acidification and elevated temperature negatively affect recruitment, oxygen  
673 consumption and calcification of the reef-building *Dendropoma cristatum* early life  
674 stages: Evidence from a manipulative field study. *Science of The Total Environment*,  
675 693, 133476.

676 Blunden, J., Derek, S.A., Hartfield, G., 2018. State of the climate in 2017. *Bull. Am.*  
677 *Meteorol. Soc.* 99.

678 Booth, C.E., McDonald, D.G., Walsh, P.J., 1984. Acid-base balance in the sea mussel,  
679 *Mytilus edulis*. II: effects of hypoxia and air-exposure on hemolymph acid-base



680 status. *Mar. Biol. Lett.* 5, 359-369.

681 Burnett, L.E., 1988. Physiological responses to air exposure: acid-base balance and  
682 the role of branchial water stores. *Am. Zool.* 28, 125-135.

683 Byrne, M., Foo, S.A., Ross, P.M., Putnam, H.M., 2020. Limitations of cross- and  
684 multigenerational plasticity for marine invertebrates faced with global climate  
685 change. *Global Change Biol.* 26, 80-102.

686 Caldeira, K., Wickett, M.E., 2003. Oceanography: anthropogenic carbon and ocean pH.  
687 *Nature* 425, 365.

688 Clements, J.C., Darrow, E.S., 2018. Eating in an acidifying ocean: a quantitative review  
689 of elevated CO<sub>2</sub> effects on the feeding rates of calcifying marine invertebrates.  
690 *Hydrobiologia* 820, 1-21.

691 Cross, E.L., Harper, E.M., Peck, L.S., 2019. Thicker shells compensate extensive  
692 dissolution in brachiopods under future ocean acidification. *Environmental  
693 Science & Technology*, 53(9), 5016–5026.

694 Dudgeon D., Morton B., 1982. Coral associated Mollusca of Tolo Harbour, Hong Kong.  
695 Morton B., Tseng C.K. (Eds.), *Proceedings of the First International Marine  
696 Biological Workshop: The Marine Flora and Fauna of Hong Kong and southern  
697 China*, Hong Kong, Hong Kong University Press, Hong Kong, pp. 627-650.

698 Ekstrom, J.A., Suatoni, L., Cooley, S.R., Pendleton, L.H., Waldbusser, G.G., Cinner, J.E.,  
699 Ritter, J., Langdon, C., Hoodonk, R.V., Gledhill, D., Wellman, K., Beck, M.W.,  
700 Brander, L.M., Rittschof, D., Doherty, C., Edwards, P., Portela, R., 2015.  
701 Vulnerability and adaptation of US shellfisheries to ocean acidification. *Nat. Clim.*

702 Change 3, 207-214.

703 Feely, R.A., Sabine, C.L., Hernandez-Ayon, J.M., Ianson, D., Hales, B., 2008. Evidence  
704 for up-welling of corrosive “acidified” water onto the continental shelf. *Science*  
705 320, 1490-1492.

706 Fitzter, S.C., Caldwell, G.S., Phoenix, V.R., Kamenos, N.A., 2014. Ocean acidification  
707 reduces the crystallographic control in juvenile mussel shells. *J. Struct. Biol.* 188,  
708 39-45.

709 Frieder, C.A., Gonzalez, J.P., Levin, L.A., 2014. Uranium in larval shells as a barometer  
710 of molluscan ocean acidification exposure. *Environ. Sci. Technol.* 48, 6401-6408.

711 Garilli, V., Rodolfo-Metalpa, R., Scuderi, D., Brusca, L., Parrinello, D., Rastrick, S.P.S.,  
712 Foggo, A., Twitchett, R.J., Hall-Spencer, J.M., Milazzo, M., 2015. Physiological  
713 advantages of dwarfing in surviving extinctions in high-CO<sub>2</sub> oceans. *Nature*  
714 *Climate Change*, 5(7), 678–682.

715 Gillikin, D.P., Lorrain, A., Meng, L., Dehairs, F., 2007. A large metabolic carbon  
716 contribution to the  $\delta^{13}\text{C}$  record in marine aragonitic bivalve shells. *Geochim.*  
717 *Cosmochim. Acta* 71, 2936-2946.

718 Gutknecht, J., Bisson, M.J., Tosteson, F.C., 1977. Diffusion of carbon dioxide through  
719 lipid bilayer membranes: effects of carbonic anhydrase, bicarbonate, and  
720 unstirred layers. *J. Gen. Physiol.* 55, 1-17.

721 Hahn, S., Rodolfo-Metalpa, R., Griesshaber, E., Schmahl, W.W., Buhl, D., Hall-Spencer,  
722 J.M., Baggini, C., Fehr, K.T., Immenhauser, A., 2012. Marine bivalve shell  
723 geochemistry and ultrastructure from modern low pH environments:

724 Environmental effect versus experimental bias. *Biogeosciences*, 9(5), 1897–1914.

725 Hall-Spencer, J.M., Rodolfo-Metalpa, R., Martin, S., Ransome, E., Fine, M., Turner,  
726 S.M., Rowley, S.J., Tedesco, D., Buia, M.C., 2008. Volcanic carbon dioxide vents  
727 show ecosystem effects of ocean acidification. *Nature* 454, 96-99.

728 Hall-Spencer, J.M., Belfiore, G., Tomatsuri, M., Porzio, L., Harvey, B.P., Agostini, S., Kon,  
729 K., 2022. Decreased diversity and abundance of marine invertebrates at CO<sub>2</sub>  
730 seeps in warm-temperate Japan. *Zoological Science* 39, 41-51.

731 Harvey, B.P., Agostini, S., Wada, S., Inaba, K., Hall-Spencer, J.M., 2018. Dissolution:  
732 the Achilles's hell of the triton shell in an acidifying ocean. *Front. Mar. Sci.* 5, 371.

733 Harvey, B.P., Allen, R., Agostini, S., Hoffmann, L.J., Kon, K., Summerfield, T.C., Wada, S.,  
734 Hall-Spencer, J.M., 2021a. Feedback mechanisms stabilise degraded turf algal  
735 systems at a CO<sub>2</sub> seep site. *Communications Biol.* 4, 219.

736 Harvey, B.P., Kerfahi, D., Jung, Y., Shin, J.-H., Adams, J.M., Hall-Spencer, J.M., 2020.  
737 Ocean acidification alters bacterial communities on marine plastic debris. *Mar.*  
738 *Pollut. Bull.* 161, 111749.

739 Harvey, B.P., Kon, K., Agostini, S., Wada, S., Hall-Spencer, J.M., 2021b. Ocean  
740 acidification locks algal communities in a species-poor early successional stage.  
741 *Global Change Biol.* 27, 2174-2187.

742 Harvey, B.P., McKeown, N.J., Rastrick, S.P.S., Bertolini, C., Foggo, A., Graham, H., Hall-  
743 Spencer, J.M., Milazzo, M., Shaw, P.W., Small, D.P., Moore, P.J., 2016. Individual  
744 and population-level responses to ocean acidification. *Scientific Reports*, 6(1),  
745 20194.

746 He, G., Xiong, X., Peng, Y., Yang, C., Xu, X., Liu, X., Liang, J., Masanja, F., Yang, K., Xu, X.,  
747 Zheng, Z., Deng, Y., Leung, J.Y.S., Zhao, L., 2023. Transcriptomic responses reveal  
748 impaired physiological performance of the pearl oyster following repeated  
749 exposure to marine heatwaves. *Sci. Total Environ.* 854, 158726.

750 Heinemann, A., Fietzke, J., Melzner, F., Bohm, F., Thomsen, J., Garbe-Schonberg, D.,  
751 Eisenhauer, A., 2012. Conditions of *Mytilus edulis* extracellular body fluids and  
752 shell composition in a pH-treatment experiment: acid-base status, trace  
753 elements and  $\delta^{11}\text{B}$ . *G-cubed* 13.

754 Higo, S., Callomon, P. and Goto, Y., 1999. Catalogue and Bibliography of the Marine  
755 Shell-Bearing Mollusca of Japan. Gastropoda. Bivalvia. Polyplacophora.  
756 Scaphopoda, *Elle Sci. Publ.*

757 Hughes, T.P., Barnes, M.L., Bellwood, D.R., Cinner, J.E., Cumming, G.S., Jackson, J.B.C.,  
758 Kleypas, J., van de Leemput, I.A., Lough, J.M., Morrison, T.H., Palumbi, S.R., van  
759 Nes, E.H., Scheffer, M., 2017. Coral reefs in the Anthropocene. *Nature* 546, 82–  
760 90.

761 Hüning, A.K., Melzner, F., Thomsen, J., Gutowska, M.A., Krämer, L., Frickenhaus, S.,  
762 Rosenstiel, P., Pörtner, H.O., Philipp, E.E.R., Lucassen, M., 2013. Impacts of  
763 seawater acidification on mantle gene expression patterns of the Baltic Sea blue  
764 mussel: implications for shell formation and energy metabolism. *Mar. Biol.* 160,  
765 1845-1861.

766 Inoue, S., Kayanne, H., Yamamoto, S., Kurihara, H., 2013. Spatial community shift  
767 from hard to soft corals in acidified water. *Nat. Clim. Chang.* 3, 683-687.

768 IPCC, 2013. Summary for policymakers. In: Stocker, T.F., Qin, D., Plattner, G.-K., Tignor,  
769 M., Allen, S.K., Boschung, J., Nauels, A., Xia, Y., Bex, V., Midgley, P.M. (Eds.),  
770 Climate Change 2013: the Physical Science Basis. Contribution of Working Group  
771 I to the Fifth Assessment Report of the Intergovernmental Panel on Climate  
772 Change. Cambridge University Press, Cambridge, United Kingdom and New York,  
773 NY, USA, pp. 1-30.

774 Kerfahi, D., Harvey, B.P., Agostini, S., Kon, K., Huang, R., Adams, J.M., Hall-Spencer,  
775 J.M., 2020. Responses of intertidal bacterial biofilm communities to increasing  
776  $p\text{CO}_2$ . *Mar. Biotechnol.* 22, 727-738.

777 Lee, T.H., McGill, R.A.R., Fitzner, S., 2021. Effects of extra feeding combined with ocean  
778 acidification and increased temperature on the carbon isotope values ( $\delta^{13}\text{C}$ ) in  
779 the mussel shell. *J. Exp. Mar. Biol. Ecol.* 541, 151562.

780 Leung, J.Y.S., Chen, Y., Nagelkerken, I., Zhang, S., Xie, Z., Connell, S.D., 2020a.  
781 Calcifiers can adjust shell building at the nanoscale to resist ocean acidification.  
782 *Small* 16, 2003186.

783 Leung, J.Y.S., Doubleday, Z.A., Nagelkerken, I., Chen, Y., Xie, Z., Connell, S.D., 2019.  
784 How calorie-rich food could help marine calcifiers in a  $\text{CO}_2$ -rich future. *Proc. R.*  
785 *Soc. B* 286, 20190757.

786 Leung, J.Y.S., Nagelkerken, I., Russell, B.D., Ferreira, C.M., Connell, S. D., 2018.  
787 Boosted nutritional quality of food by  $\text{CO}_2$  enrichment fails to offset energy  
788 demand of herbivores under ocean warming, causing energy depletion and  
789 mortality. *Science of the Total Environment*, 639, 360–366.

790 Leung, J.Y.S., Russell, B.D., Connell, S.D., 2017. Mineralogical plasticity acts as a  
791 compensatory mechanism to the impacts of ocean acidification. *Environmental*  
792 *Science & Technology*, 51(5), 2652–2659.

793 Leung, J.Y.S., Russell, B.D., Connell, S.D., 2020. Linking energy budget to physiological  
794 adaptation: how a calcifying gastropod adjusts or succumbs to ocean  
795 acidification and warming. *Sci. Total Environ.* 715, 136939.

796 Leung, J.Y.S., Zhang, S., Connell, S.D., 2022. Is ocean acidification really a threat to  
797 marine calcifiers? A systematic review and meta-analysis of 980+ studies  
798 spanning two decades. *Small* 18, 2107407.

799 Levin, L.A., Hönisch, B., Frieder, C.A., 2015. Geochemical proxies for estimating faunal  
800 exposure to ocean acidification. *Geochemistry* 28, 62-73.

801 Li, S., Liu, C., Huang, J., Liu, Y., Zhang, S., Zheng, G., Xie, L., Zhang, R., 2016.  
802 Transcriptome and biomineralization responses of the pearl oyster *Pinctada*  
803 *fuscata* to elevated  $p\text{CO}_2$  and temperature. *Sci. Rep.* 6, 18493.

804 Liu, Y.W., Aciego, S.M., Wanamaker, A.D., 2015. Environmental controls on the boron  
805 and strontium isotopic composition of aragonite shell material of cultured  
806 *Arctica islandica*. *Biogeosciences* 12, 3351-3368.

807 Liu, Y., Sutton, J.N., Ries, J.B., Eagle, R.A., 2020. Regulation of calcification site pH is a  
808 polyphyletic but not always governing response to ocean acidification. *Sci. Adv.*  
809 6, eaax1314.

810 Lu, Y., Wang, L., Wang, L., Cong, Y., Yang, G., Zhao, L., 2018. Deciphering carbon  
811 sources of mussel shell carbonate under experimental ocean acidification and

812 warming. Mar. Environ. Res. 142, 141-146.

813 Lucas, A., Beninger, P.G., 1985. The use of physiological condition indices in marine  
814 bivalve aquaculture. Aquaculture 44, 187-200.

815 Lucey, N.M., Lombardi, C., DeMarchi, L., Schulze, A., Gambi, M.C., Calosi, P., 2015. To  
816 brood or not brood: are marine invertebrates that protect their offspring more  
817 resilient to ocean acidification? Sci. Rep. 5, 12009.

818 Marin, F., Le Roy, N., Marie, B., 2012. The formation and mineralization of mollusk  
819 shell. Front. Biosci. S4, 1099-1125.

820 Martins, M., Carreiro-Silva, M., Martins, G.M., e Ramos, J.B., Viveiros, F., Couto, R.P.,  
821 Parra, H., Monteiro, J., Gallo, F., Silva, C., Teodosio, A., Guilini, K., Hall-Spencer,  
822 J.M., Leitao, F., Chicharo, L., Range, P., 2021. *Ervilia castanea* (Mollusca, Bivalvia)  
823 populations adversely affected at CO<sub>2</sub> seeps in the North Atlantic. Sci. Total  
824 Environ. 754, 142044.

825 McConnaughey, T.A., Burdett, J., Whelan, J.F., Paull, C.K., 1997. Carbon isotopes in  
826 biological carbonates: respiration and photosynthesis. Geochim. Cosmochim.  
827 Acta 61, 611-622.

828 Melzner, F., Mark, F.C., Seibel, B.A., Tomanek, L., 2020. Ocean acidification and  
829 coastal marine invertebrates: tracking CO<sub>2</sub> effects from seawater to the cell.  
830 Annual Rev. Marine Sci. 12, 499-523.

831 Morton, B., 2019. The biology and functional morphology of *Septifer bilocularis* and  
832 *Mytilisepta virgata* (Bivalvia: Mytiloidea) from corals and the exposed rocky  
833 shores, respectively, of Hong Kong. Regional Studies in Marine Science 25,

834 100454.

835 Morton, B., Leung, P.T.Y., Wei, J., Lee, G.Y., 2020. A morphological and genetic  
836 comparison of *Septifer bilocularis*, *Mytilisepta virgata* and *Brachidontes*  
837 *variabilis* (Bivalvia: Mytiloidea) from Hong Kong and erection of the  
838 Mytiliseptiferinae sub-fam. nov. Regional Studies in Marine Science 34, 100981.

839 Noisette, F., Comtet, T., Legrand, E., Bordeyne, F., Davoult, D., Martin, S., 2014. Does  
840 encapsulation protect embryos from the effects of ocean acidification? The  
841 example of *Crepidula fornicata*. PLoS One 9, 1-11.

842 Palmer, A.R., 1992. Calcification in marine molluscs: how costly is it? Proc. Natl. Acad.  
843 Sci. U. S. A. 89, 1379-1382.

844 Parker, L.M., O'Connor, W.A., Raftos, D.A., Pörtner, H.O., Ross, P.M., 2015. Persistence  
845 of positive carryover effects in oysters following transgenerational exposure to  
846 ocean acidification. PloS One 10, e0132276.

847 Pierrot, D.E., Lewis, E., Wallace, D.W.R., 2006. MS Excel Program Developed for CO<sub>2</sub>  
848 System Calculations, ORNL/CDIAC-105a, Carbon Dioxide Information Analysis  
849 Center. Oak Ridge National Laboratory, U.S. Department of Energy, Oak Ridge,  
850 Tennessee.

851 Rajan, K.C., Meng, Y., Yu, Z., Roberts, S.B., Vengatesen, T., 2021. Oyster  
852 biomineralization under ocean acidification: from genes to shell. Glob. Chang.  
853 Biol. 27, 3779-3797.

854 Ramajo, L., Pérez-Léon, E., Hendriks, I.E., Marbà, N., Krause-Jensen, D., Sejr, M.K.,  
855 Blicher, M.E., Lagos, N.A., Olsen, Y.S., Duarte, C.M., 2016. Food supply confers



856 calcifiers resistance to ocean acidification. *Sci. Rep.* 6, 19374.

857 Ramesh, K., Yarra, T., Clark, M.S., John, U., Melzner, F., 2019. Expression of  
858 calcification- related ion transporters during blue mussel larval development.  
859 *Ecol. Evol.* 9, 7157-7172.

860 Rastrick, S.S.P., Graham, H., Azetsu-Scott, K., Calosi, P., Chierici, M., Fransson, A., et al.,  
861 2018. Using natural analogues to investigate the effects of climate change and  
862 ocean acidification on northern ecosystems. *ICES J. Mar. Sci.* 75, 2299-2311.

863 Récluz C.A., 1848. Description d'un nouveau genre de coquille bivalve nommé  
864 Septifère (Septifer). *Rev. Zool. Soc. Cuv.* 11, 275-279

865 Reusch, T.B.H., 2014. Climate change in the oceans: evolutionary versus  
866 phenotypically plastic responses of marine animals and plants. *Evol. Appl.* 7,  
867 104-122.

868 Ricevuto, E., Vizzini, S., Gambi, M.C., 2015. Ocean acidification effects on stable  
869 isotope signatures and trophic interactions of polychaete consumers and  
870 organic matter sources at a CO<sub>2</sub> shallow vent system. *Journal of Experimental*  
871 *Marine Biology and Ecology* 468, 105-117.

872 Ross, P.M., Parker, L., Byrne, M., 2016. Transgenerational responses of molluscs and  
873 echinoderms to changing ocean conditions. *ICES J. Mar. Sci.* 73, 537-549.

874 Saleuddin, A.S.M., Petit H.P., 1983. The mode of formation and the structure of the  
875 periostracum. In *Physiology, Part 1 The Mollusca*, vol. 4 (eds. Saleuddin, A.S.M.,  
876 Wilbur, K.M.). Academic Press, New York, pp. 199-234.

877 Sheridan, J.A., Bickford, D., 2011. Shrinking body size as an ecological response to

878 climate change. *Nat. Clim. Change* 1, 401-406.

879 Sillanpää, J.K., Sundh, H., Sundell, K.S., 2018. Calcium transfer across the outer  
880 mantle epithelium in the Pacific oyster, *Crassostrea gigas*. *Proc. R. Soc. B* 285,  
881 20181676.

882 Spalding, C., Finnegan, S., Fischer, W.W., 2017. Energetic costs of calcification under  
883 ocean acidification. *Global Biogeochem. Cycles* 31, 866-877.

884 Thomsen, J., Stapp, L.S., Haynert, K., Schade, H., Danelli, M., Lannig, G., Wegner, K.M.,  
885 Melzner, F., 2017. Naturally acidified habitat selects for ocean acidification-  
886 tolerant mussels. *Sci. Adv.* 3, e1602411.

887 Waldbusser, G.G., Brunner, E.L., Haley, B.A., Hales, B., Langdon, C.J., Pahl, F.G., 2013.  
888 A developmental and energetic basis linking larval oyster shell formation to  
889 acidification sensitivity. *Geophys. Res. Lett.* 40, 2171-2176.

890 Waldbusser, G.G., Hales, B., Langdon, C.J., Haley, B.A., Schrader, P., Brunner, E.L., Gray,  
891 M.W., Miller, C.A., Gimenez, I., 2014. Saturation-state sensitivity of marine  
892 bivalve larvae to ocean acidification. *Nat. Clim. Change* 5, 273-280.

893 Waldbusser, G.G., Gray, M.W., Hales, B., Langdon, C.J., Haley, B.A., Gimenez, I., Smith,  
894 S.R., Brunner, E.L., Hutchinson, G., 2016. Slow shell building, a possible trait for  
895 resistance to the effects of acute ocean acidification. *Limnol. Oceanogr.* 61,  
896 1969-1983.

897 Witkowski, C. R., Agostini, S., Harvey, B. P., van der Meer, M. T. J., Sinninghe Damsté, J.  
898 S., Schouten, S., 2019. Validation of carbon isotope fractionation in algal lipids as  
899 a  $p\text{CO}_2$  proxy using a natural  $\text{CO}_2$  seep (Shikine Island, Japan). *Biogeosciences*,

900 16, 4451–4461.

901 Wikelski, M., Thom, C., 2000. Marine iguanas shrink to survive El Nino. *Nature* 403,  
902 37-38.

903 Zhao, L., Liu, B., An, W., Deng, Y., Lu, Y., Liu, B., Wang, L., Cong, Y., Sun, X., 2019a.  
904 Assessing the impact of elevated  $p\text{CO}_2$  within and across generations in a highly  
905 invasive fouling mussel (*Musculista senhousia*). *Sci. Total Environ.* 689, 322-333.

906 Zhao, L., Lu, Y., Yang, F., Liang, J., Deng, Y., 2020b. Transgenerational biochemical  
907 effects of seawater acidification on the Manila clam (*Ruditapes philippinarum*).  
908 *Sci. Total Environ.* 710, 136420.

909 Zhao, L., Milano, S., Walliser, E.O., Schöne, B.R., 2018b. Bivalve shell formation in a  
910 naturally  $\text{CO}_2$ -enriched habitat: unraveling the resilience mechanisms from  
911 elemental signatures. *Chemosphere* 203, 132-138.

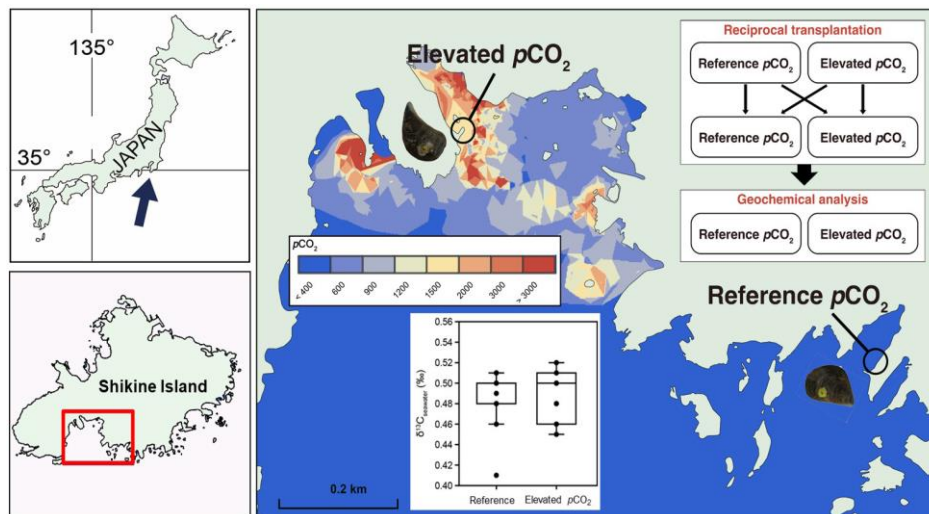
912 Zhao, L., Schöne, B.R., Mertz-Kraus, R., Yang, F., 2017a. Insights from sodium into the  
913 impacts of elevated  $p\text{CO}_2$  and temperature on bivalve shell formation. *J. Exp.*  
914 *Mar. Biol. Ecol.* 486, 148-154.

915 Zhao, L., Schöne, B.R., Mertz-Kraus, R., Yang, F., 2017b. Sodium provides unique  
916 insights into transgenerational effects of ocean acidification on bivalve shell  
917 formation. *Sci. Total Environ.* 577, 360-366.

918 Zhao, L., Shirai, K., Tanaka, K., Milano, S., Higuchi, T., Murakami-Sugihara, N., Walliser,  
919 E.O., Yang, F., Deng, Y., Schöne, B.R., 2020a. A review of transgenerational  
920 effects of ocean acidification on marine bivalves and their implications for  
921 sclerochronology. *Estuar. Coast. Shelf S.*, 235, 106620.

- 922 Zhao, L., Shirai, K., Murakami-Sugihara, N., Higuchi, T., Tanaka, K., 2019b. Mussel  
923 periostracum as a high-resolution archive of soft tissue  $\delta^{13}\text{C}$  records in coastal  
924 ecosystems. *Geochem. Cosmochim. Acta* 260, 232-243.
- 925 Zhao, L., Yang, F., Milano, S., Han, T., Walliser, E.O., Schöne, B.R., 2018a.  
926 Transgenerational acclimation to seawater acidification in the Manila clam  
927 *Ruditapes philippinarum*: preferential uptake of metabolic carbon. *Sci. Total*  
928 *Environ.* 627, 95-103.
- 929 Zhao, L., Yang, F., Yan, X., 2013. Stable isotopes and fatty acids as dietary tracers of  
930 intertidal bivalves. *Fisheries Sci.* 79, 749-756.

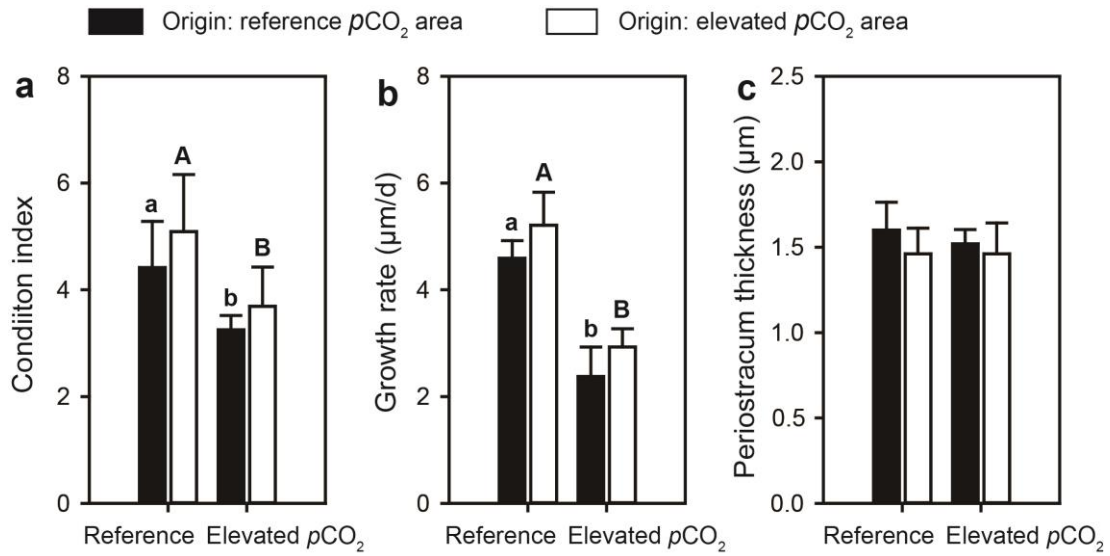
931



932

933 **Figure 1** Volcanic CO<sub>2</sub> seeps off Shikine Island (Japan) showing the spatial variability  
934 in seawater pCO<sub>2</sub> (based on data by Agostini et al. 2018 using ArcGIS 10.2 software)  
935 and the average δ<sup>13</sup>C ratio of seawater DIC near and outside the CO<sub>2</sub> seeps (based on  
936 data collected during the survey in October 2019). Experimental design showing  
937 reciprocal transplantation and high-resolution analysis of shell geochemistry of  
938 mussels (*Septifer bilocularis*) collected from the reference and elevated pCO<sub>2</sub> sites.  
939 Mussels were labelled with a reference number using queen bee marking tags for  
940 identification.

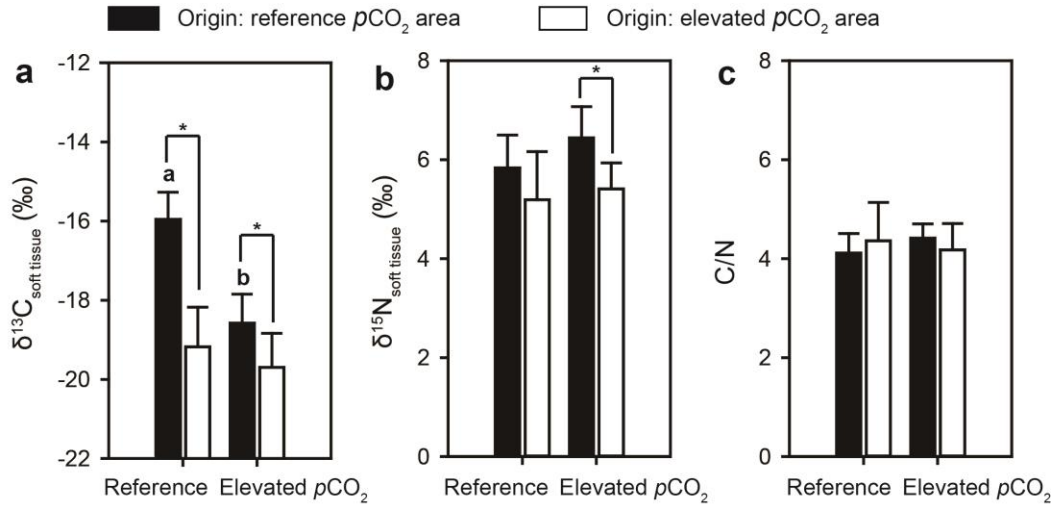
941



942

943 **Figure 2** Growth performance of mussels collected from the reference  $p\text{CO}_2$  and  
 944 elevated  $p\text{CO}_2$  areas following reciprocal transplantation. (a) condition index; (b)  
 945 growth rate; (c) the thickness of periostracum. Solid black fill indicates that  
 946 individuals originated from reference  $p\text{CO}_2$  area, and solid white fill indicates that  
 947 individuals originated from elevated  $p\text{CO}_2$  area. Small letters indicate a significant  
 948 difference in mussels originating from the reference  $p\text{CO}_2$  area under reference and  
 949 elevated  $p\text{CO}_2$  conditions, and capital letters indicates significant differences in those  
 950 mussels originally collected from the elevated  $p\text{CO}_2$  area.

951



952

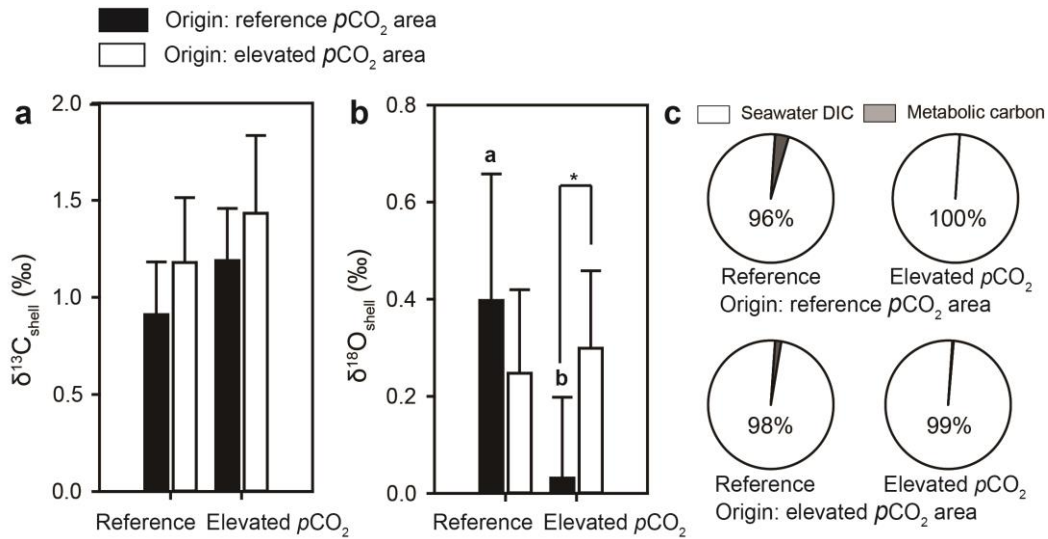
953 **Figure 3** Somatic stable isotopic signatures of mussels from the reference  $p\text{CO}_2$  and  
 954 elevated  $p\text{CO}_2$  areas following reciprocal transplantation. (a) stable carbon isotope  
 955 ( $\delta^{13}\text{C}$ ); (b) stable nitrogen isotope ( $\delta^{15}\text{N}$ ); (c) C/N. Small letters indicate a significant  
 956 difference in mussels collected from the reference  $p\text{CO}_2$  area under low and high  
 957  $p\text{CO}_2$  conditions, and an asterisk indicates a significant difference between  
 958 reciprocally transplanted mussels subjected to the same  $p\text{CO}_2$  scenario.

959

960

961

962

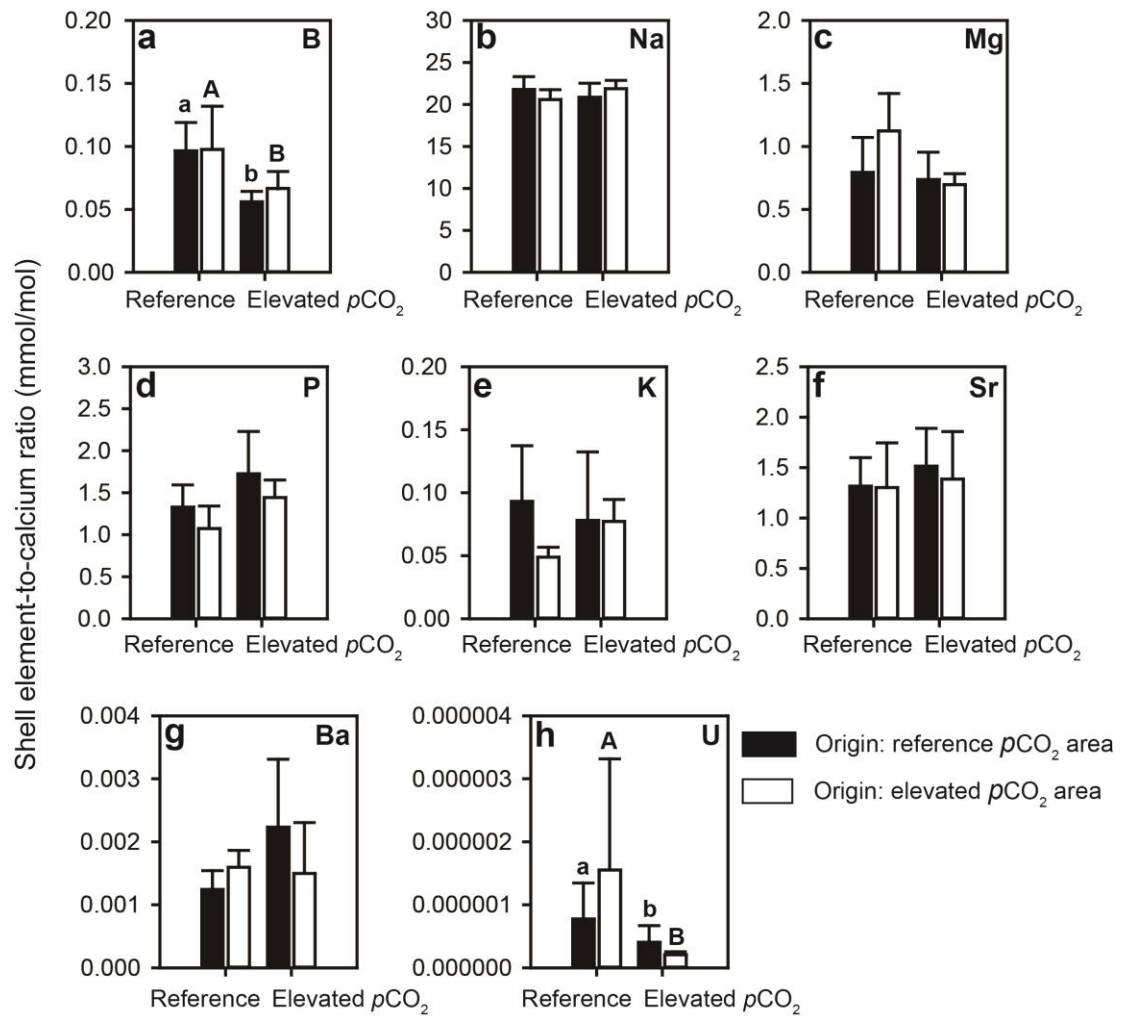


963

964 **Figure 4** Shell geochemical properties of mussels from the reference  $pCO_2$  and  
 965 elevated  $pCO_2$  areas following reciprocal transplantation. (a) stable carbon isotope  
 966 ( $\delta^{13}C$ ); (b) stable oxygen isotope ( $\delta^{18}O$ ); (c) relative contribution of seawater DIC and  
 967 metabolic carbon to shell carbonate. Small letters indicate a significant difference in  
 968 mussels collected from the reference  $pCO_2$  area under low and high  $pCO_2$  conditions,  
 969 and an asterisk indicates a significant difference between reciprocally transplanted  
 970 mussels subjected to the same  $pCO_2$  scenario.

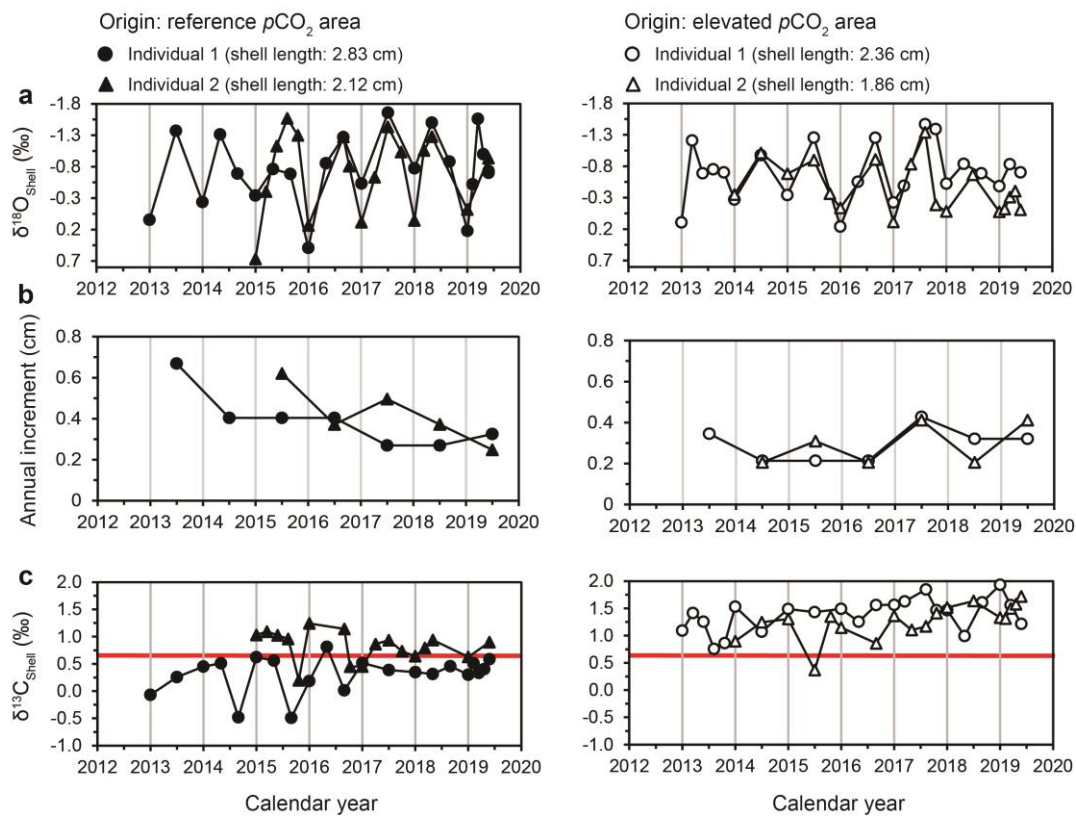
971





972

973 **Figure 5** Shell element-to-calcium ratios of mussels collected from the reference  
 974 *p*CO<sub>2</sub> and elevated *p*CO<sub>2</sub> areas following reciprocal transplantation. (a) B/Ca; (b)  
 975 Na/Ca; (c) Mg/Ca; (d) P/Ca; (e) K/Ca; (f) Sr/Ca; (g) Ba/Ca; (h) U/Ca. Small letters  
 976 indicate a significant difference in mussels originating from the reference *p*CO<sub>2</sub> area  
 977 under reference and elevated *p*CO<sub>2</sub> conditions, and capital letters indicates  
 978 significant differences in those mussels originally collected from the elevated *p*CO<sub>2</sub>  
 979 area.



980

981 **Figure 6** Analysis of shell geochemistry in mussels collected from the reference  $p\text{CO}_2$

982 and elevated  $p\text{CO}_2$  areas. Two individuals were selected from the reciprocal

983 transplant experiment for further analysis. This used only those procedural control

984 individuals that had been collected from, and transplanted back into the same site. (a)

985 stable oxygen isotope ( $\delta^{18}\text{O}$ ) of mussel shells according to which each shell portion

986 was placed into a temporal context and ontogenetic age of each mussel was then

987 determined; (b) annual increment calculated according to the temporal aligned shell

988 portion; (c) stable carbon isotope ( $\delta^{13}\text{C}$ ) of mussel shells. Red lines indicate the mean

989  $\delta^{13}\text{C}$  ratio of seawater DIC in the reference  $p\text{CO}_2$  and elevated  $p\text{CO}_2$  areas.